



北京大学 物理学院
PEKING UNIVERSITY

走向核素版图的边缘

Towards the edge of nuclear landscape

许甫荣

I. 核物理发展历史 (重要里程碑)

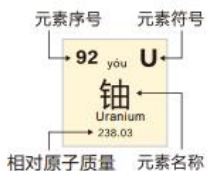
II. 当前核物理前沿



元素周期表

IA																											0	
1 qīng H 氢 Hydrogen 1.0079	IIA																2 hāi He 氦 Helium 4.002602											
3 lì Li 锂 Lithium 6.941	4 gāi Be 铍 Beryllium 9.0122																	5 péng B 硼 Boron 10.811	6 tàn C 碳 Carbon 12.0107	7 dàn N 氮 Nitrogen 14.0067	8 yǎng O 氧 Oxygen 15.9994	9 fú F 氟 Fluorine 18.9984032	10 nǚ Ne 氖 Neon 20.1797					
11 nǎ Na 钠 Sodium 22.9898	12 měi Mg 镁 Magnesium 24.305																	13 lǚ Al 铝 Aluminum 26.9815386	14 guā Si 硅 Silicon 28.0855	15 pǐ P 磷 Phosphorus 30.973762	16 liú S 硫 Sulfur 32.065	17 lǜ Cl 氯 Chlorine 35.453	18 yì Ar 氩 Argon 39.948					
																	VIII										IB	IIB
19 jiǎ K 钾 Potassium 39.098	20 gāi Ca 钙 Calcium 40.08	21 kāng Sc 钪 Scandium 44.956	22 tǎi Ti 钛 Titanium 47.867	23 fān V 钒 Vanadium 50.9415	24 gā Cr 铬 Chromium 51.9961	25 mǎng Mn 锰 Manganese 54.938045	26 tiě Fe 铁 Iron 55.845	27 gǔ Co 钴 Cobalt 58.933195	28 niè Ni 镍 Nickel 58.6934	29 tóng Cu 铜 Copper 63.546	30 zin Zn 锌 Zinc 65.38	31 jiǎ Ga 镓 Gallium 69.723	32 zhē Ge 锗 Germanium 72.64	33 zhēn As 砷 Arsenic 74.9216	34 xī Se 硒 Selenium 78.96	35 xiǔ Br 溴 Bromine 79.904	36 kǐ Kr 氪 Krypton 83.798											
37 rú Rb 铷 Rubidium 85.467	38 sī Sr 锶 Strontium 87.62	39 yī Y 钇 Yttrium 88.906	40 gāo Zr 锆 Zirconium 91.22	41 ní Nb 铌 Niobium 92.90638	42 mǒ Mo 钼 Molybdenum 95.96	43 dé Tc 锝 Technetium 97.9072	44 kǎo Ru 钌 Ruthenium 101.07	45 iáo Rh 铑 Rhodium 102.90550	46 bǎ Pd 钯 Palladium 106.42	47 yín Ag 银 Silver 107.8682	48 gē Cd 镉 Cadmium 112.411	49 yīn In 铟 Indium 114.818	50 xī Sn 锡 Tin 118.710	51 tī Sb 锑 Antimony 121.760	52 dié Te 碲 Tellurium 127.60	53 diǎn I 碘 Iodine 126.90447	54 xīn Xe 氙 Xenon 131.29											
55 cǎi Cs 铯 Cesium 132.905	56 bāi Ba 钡 Barium 137.33	La-Lu 镧系 Lanthanides		72 hā Hf 铪 Hafnium 178.49	73 tǎn Ta 钽 Tantalum 180.94788	74 wú W 钨 Tungsten 183.84	75 lǎ Re 铼 Rhenium 186.207	76 ōs Os 锇 Osmium 190.23	77 yī Ir 铱 Iridium 192.222	78 pà Pt 铂 Platinum 195.084	79 jīn Au 金 Gold 196.966569	80 gōng Hg 汞 Mercury 200.59	81 tā Tl 铊 Thallium 204.3833	82 qiān Pb 铅 Lead 207.2	83 bī Bi 铋 Bismuth 208.98040	84 pō Po 钋 Polonium 208.9824	85 ài At 砹 Astatine 208.9871	86 dǎn Rn 氡 Radon 222.0176										
87 fēng Fr 钫 Francium 223	88 lái Ra 镭 Radium 226	Ac-Lr 锕系 Actinides		104 lǎo Rf 钿 Rutherfordium (261)	105 dǔ Db 𨭎 Dubnium (262)	106 yī Sg 𨭏 Seaborgium (263)	107 lǚ Bh 𨭐 Bohrium (262)	108 shàn Hs 𨭑 Hassium (265)	109 mǎi Mt 𨭒 Meitnerium (266)	110 dǎ Ds 𨭓 Darmstadtium (269)	111 sūn Rg 𨭔 Roentgenium (272)	112 gē Cn 𨭕 Copernicium (285)	113 ní Nh 𨭖 Nihonium (284)	114 fū Fl 𨭗 Flerovium (289)	115 mǎo Mc 𨭘 Moscovium (288)	116 lì Lv 𨭙 Livermorium (293)	117 tiān Ts 𨭚 Tennessine (294)	118 ōg Og 𨭛 Oganesson (294)										
119	120																											

超重元素



57 lán La 镧 Lanthanum 138.905	58 shì Ce 铈 Cerium 140.12	59 pǐ Pr 镨 Praseodymium 140.91	60 nǎ Nd 钕 Neodymium 144.2	61 pǐ Pm 钷 Promethium 147	62 shàn Sm 钐 Samarium 150.4	63 yōu Eu 铕 Europium 151.96	64 gǎ Gd 钆 Gadolinium 157.25	65 tē Tb 铽 Terbium 158.93	66 dì Dy 镝 Dysprosium 162.5	67 huò Ho 铈 Holmium 164.93	68 ěr Er 铈 Erbium 167.2	69 dié Tm 铈 Thulium 168.93	70 yī Yb 铈 Ytterbium 173	71 lǚ Lu 铈 Lutetium 174.96
89 à Ac 锕 Actinium 227.03	90 tǎ Th 钍 Thorium 232.04	91 pǎ Pa 镤 Protactinium 231.04	92 yōu U 铀 Uranium 238.03	93 nǚ Np 镎 Neptunium 237.05	94 bā Pu 钷 Plutonium 244	95 mǎ Am 镅 Americium 243	96 jū Cm 钷 Curium 247	97 kǎi Bk 锫 Berkelium 247	98 kǎi Cf 锪 Californium 251	99 ài Es 锿 Einsteinium 254	100 fēi Fm 镄 Fermium 257	101 mǎ Md 镅 Mendelevium 258	102 nǚ No 镎 Nobelium 259	103 lǚ Lr 镎 Lawrencium 260

主族金属 副族金属 非金属元素 稀有气体 人造元素

元素周期表设计作者：袁海博

Classification of elements



Dmitri Ivanovich
Mendeleev
(1834—1907年)

EJC2011, A. Lopez-Martens

Regularities in the chemical properties of the elements

ОПЫТЪ СИСТЕМЫ ЭЛЕМЕНТОВЪ.

ОСНОВАННОЙ НА ИХЪ АТОМНОМЪ ВѢСѢ И ХИМИЧЕСКОМЪ СХОДСТВѢ.

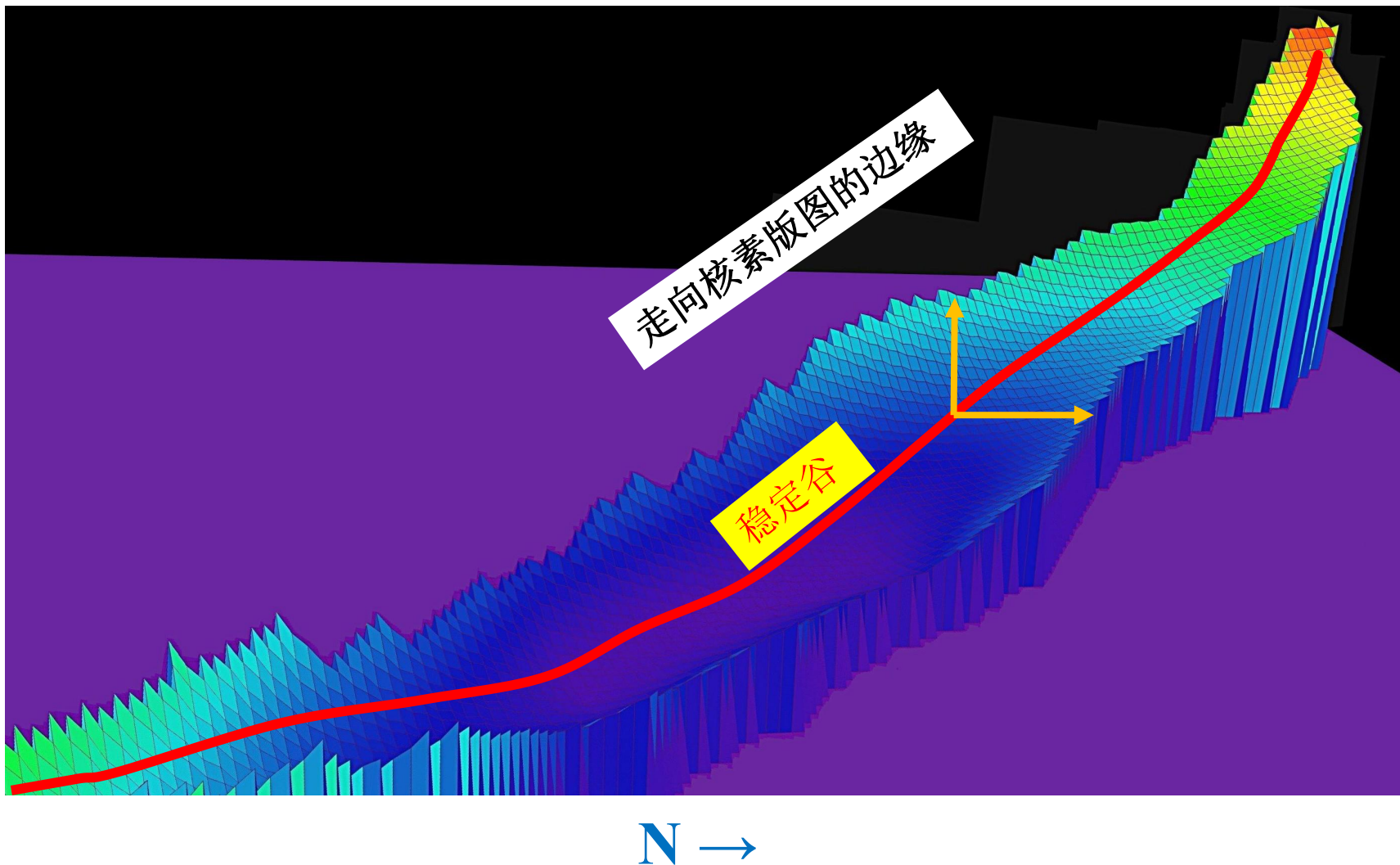
		Ti = 50	Zr = 90	? = 180.	
		V = 51	Nb = 94	Ta = 182.	
		Cr = 52	Mo = 96	W = 186.	
		Mn = 55	Rh = 104,4	Pt = 197,4.	
		Fe = 56	Ru = 104,4	Ir = 198.	
	Ni = Co = 59	Pt = 106,8	O = 199.		
H = 1		Cu = 63,4	Ag = 108	Hg = 200.	
	Be = 9,4	Mg = 24	Zn = 65,2	Cd = 112	
	B = 11	Al = 27,9	? = 68	U = 116	At = 197?
	C = 12	Si = 28	Ge? = 70	Sn = 118	
	N = 14	P = 31	As = 75	Sb = 122	Bi = 210?
	O = 16	S = 32	Se = 79,4	Te = 128?	
	F = 19	Cl = 35,5	Br = 80	I = 127	
Li = 7	Na = 23	K = 39	Rb = 85,4	Cs = 133	Tl = 204.
		Ca = 40	Sr = 87,6	Ba = 137	Pb = 207.
		Sc? = 45	Ce = 92		
		?Er = 56	La = 94		
		?Yt = 60	Di = 95		
		?In = 75,6	Th = 118?		

Д. Менделѣевъ (1869)

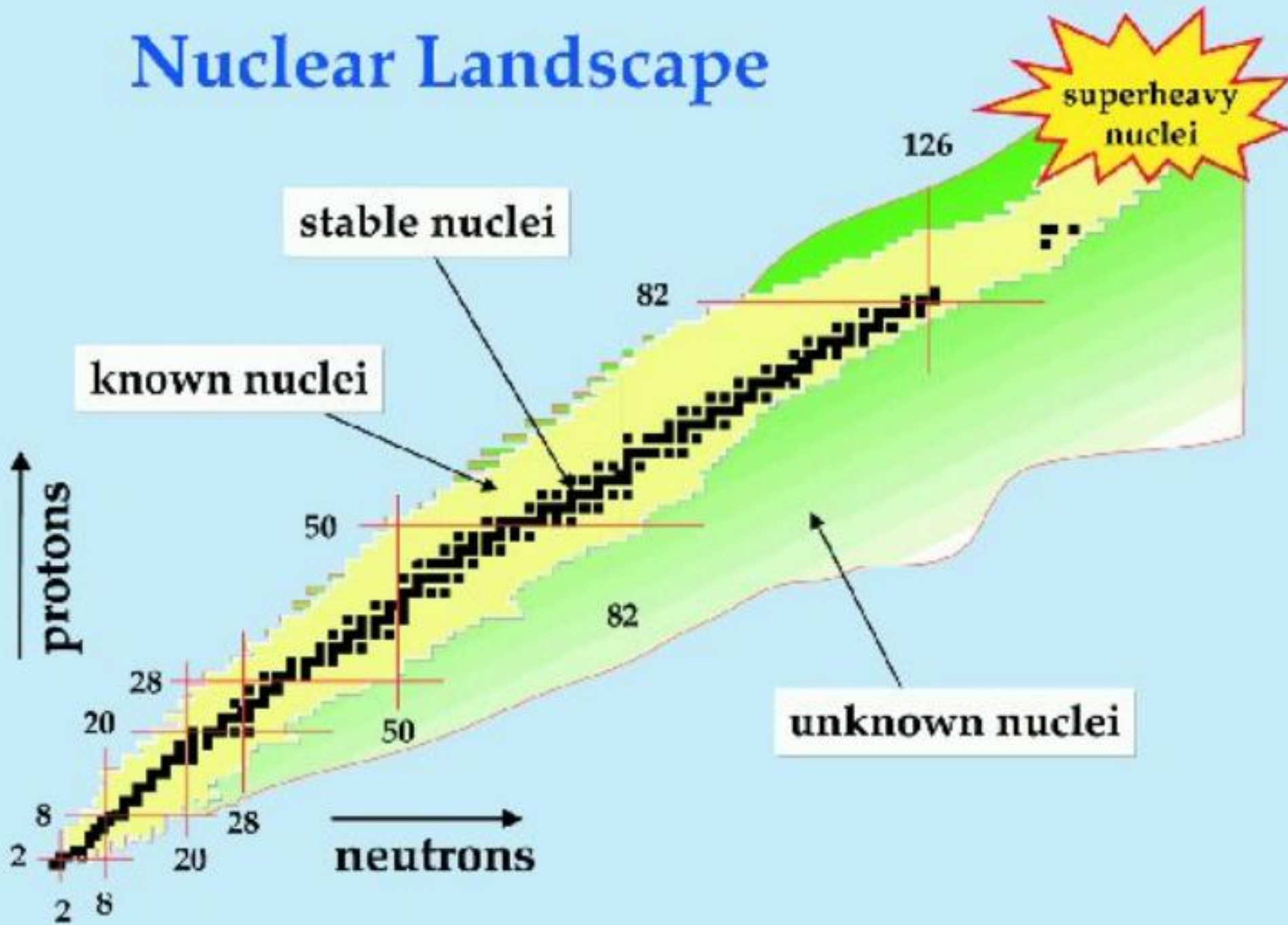
前后20年，1869年门捷列夫发现了元素周期律：元素质量+化学性质

给当时未知元素留空位，预言未知元素：比如，氢、氦、氩、氪、氙和氡

核素版图 Nuclear Landscape



Nuclear Landscape

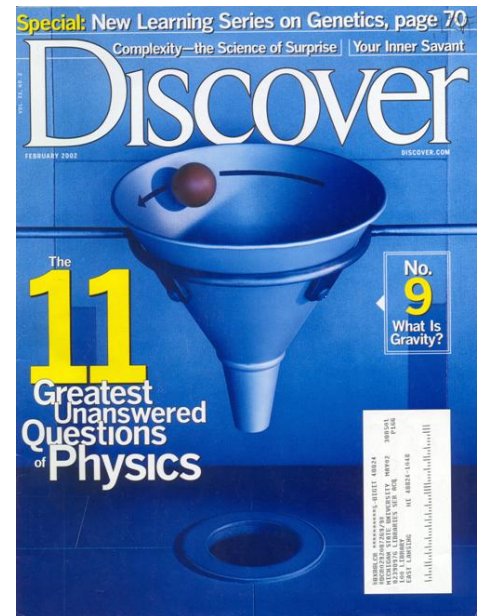


What is the Universe made of ?

- About 95% are of the dark matter and dark energy (don't know exactly what they are)
- Nuclear astrophysics attempts to study the rest 5%

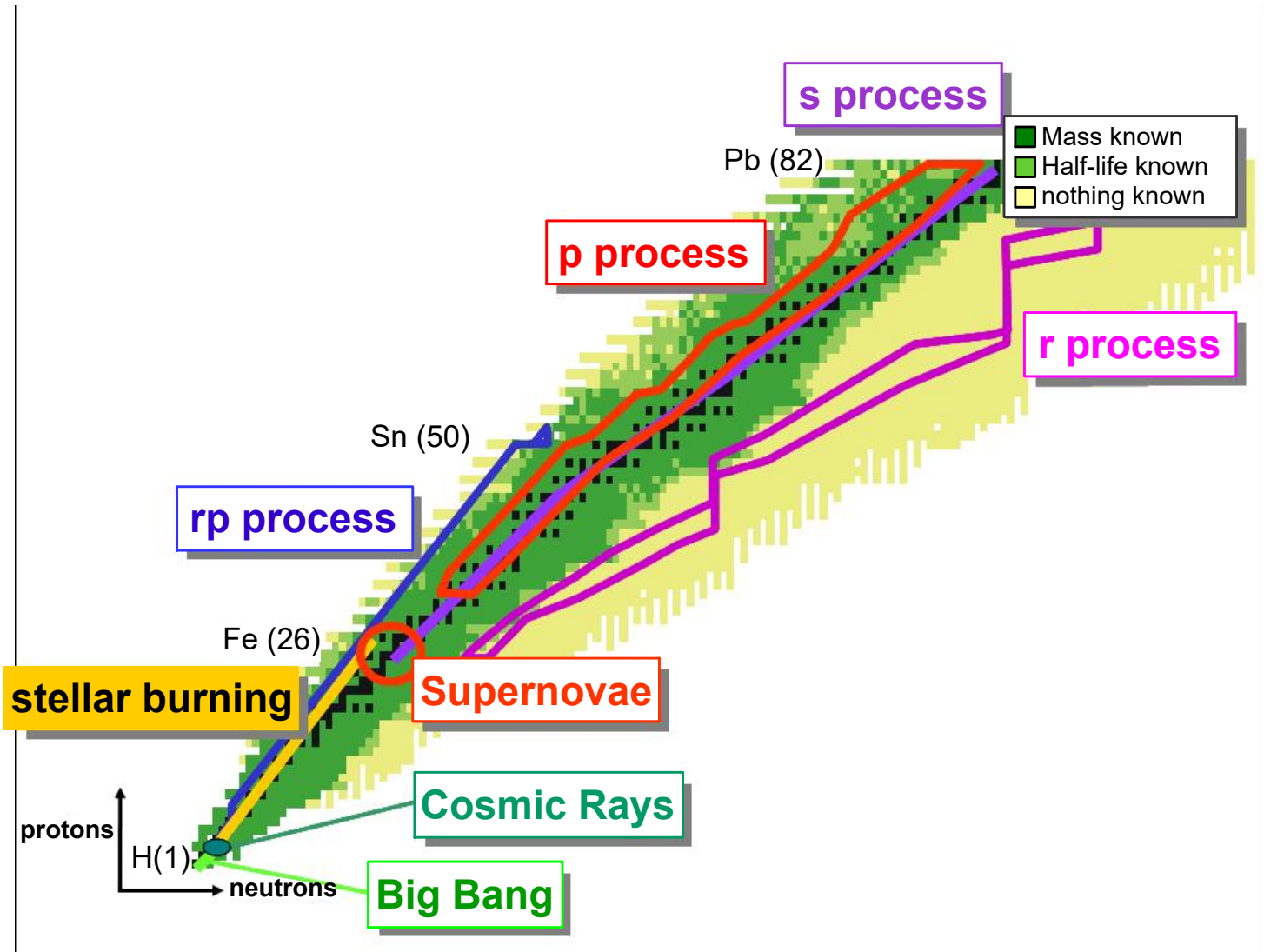
The USA National Academy of Science Report listed 11 questions of physics

- The Question # 3: How were the elements from iron to uranium made?



宇宙时间	主要事件	距今时间
~ 0	奇点,宇宙大爆炸	~ 150 亿年
10^{-43} 秒	普朗克时间,四种力统一	~ 150 亿年
10^{-6} 秒	强子时间,四种力分离	~ 150 亿年
2 秒	夸克囚禁时间,粒子产生	~ 150 亿年
3 分	辐射时间,合成氦	~ 150 亿年
1 万年	物质时间,宇宙变成物质为主	~ 150 亿年
70 万年	脱耦时间,宇宙变成透明	~ 150 亿年
~ 20 亿年	星系开始形成	~ 130 亿年
30 亿年	星系开始成团	~ 120 亿年
40 亿年	前银河系吸积	~ 110 亿年
41 亿年	第一代恒星形成	~ 109 亿年
50 亿年	第二代恒星形成	~ 100 亿年
100 亿年	第三代恒星形成,原太阳星云形成	~ 50 亿年
104 亿年	原太阳星云吸积	~ 46 亿年

核天体物理： Element production



Location of the Neutron Dripline at Fluorine and Neon

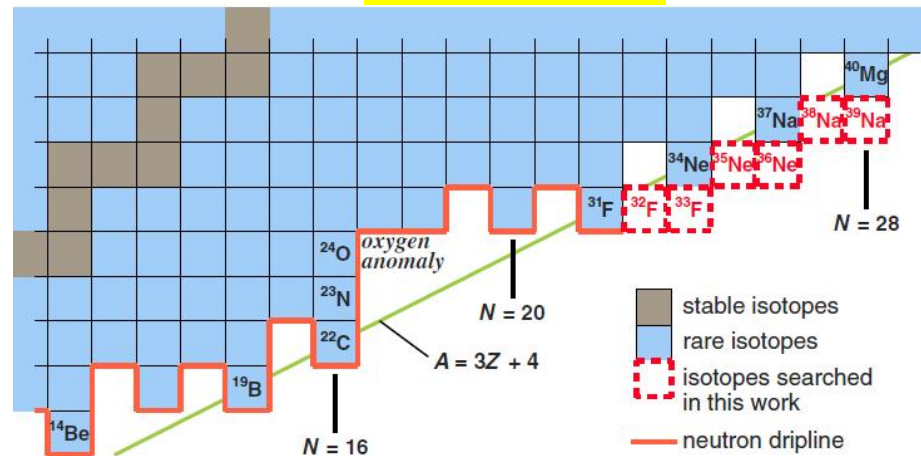
D. S. Ahn,¹ N. Fukuda,¹ H. Geissel,⁵ N. Inabe,¹ N. Iwasa,⁴ T. Kubo,^{1,*}† K. Kusaka,¹ D. J. Morrissey,⁶
 D. Murai,³ T. Nakamura,² M. Ohtake,¹ H. Otsu,¹ H. Sato,¹ B. M. Sherrill,⁶ Y. Shimizu,¹ H. Suzuki,¹
 H. Takeda,¹ O. B. Tarasov,⁶ H. Ueno,¹ Y. Yanagisawa,¹ and K. Yoshida¹

¹RIKEN Nishina Center for Accelerator-Based Science, RIKEN, 2-1 Hirosawa, Wako, Saitama 351-0198, Japan

美国物理学会选出的“2019年物理学发生的十大事件”

4. 中子滴线得到延伸

走向边缘

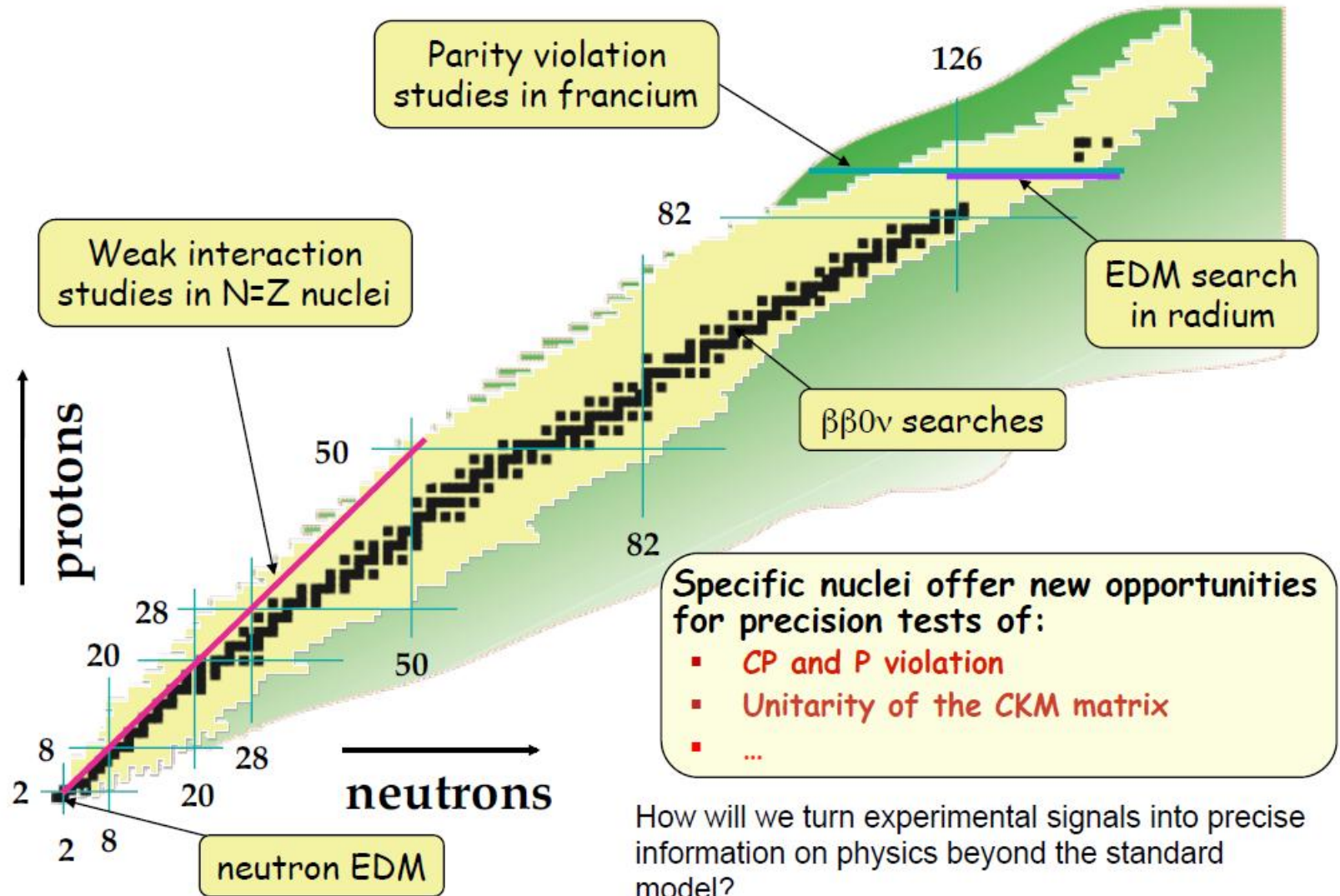


中子滴线停滞在元素周期表的前8个元素的中子滴线 (粉线) 已经20年。2019年, 日本科学家将中子滴线延伸至氟 (F) 和氖 (Ne) (绿线)

物理学家希望下一个元素的滴线不要再等上20年。下一代稀有同位素装置计划在两年内投入使用, 可能会将滴线延伸至镁元素, 即元素周期表中的第12号元素。

与粒子物理交叉

Testing the fundamental symmetries of nature



Constituents of matter



John Dalton

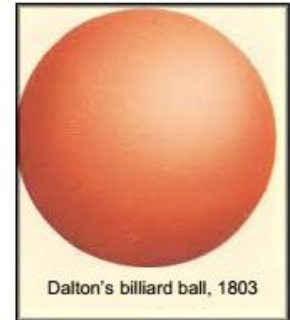
1803 :

-matter is made of **atoms**

-atoms of the same element are identical

-atoms of an element can combine with those of other elements to form compounds

-atoms of different elements have different masses



I. 核物理发展历史（重要里程碑）

At the dawn of the 20th century



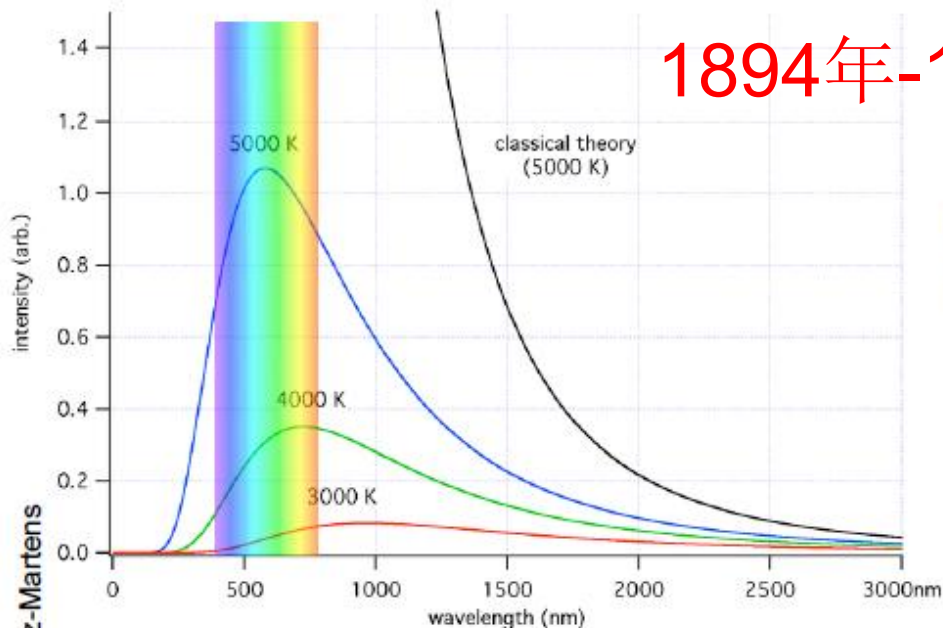
（1824-1907年）

“There is nothing new to be discovered in physics now.
All that remains is more and more precise measurement.”

William Thomson (Lord Kelvin), 1900
British Association for the advancement of Science

A few clouds in the sky.....

ultraviolet catastrophe: spectral distribution of thermal radiation from matter



1894年-1900年



Max Planck

Matter can only absorb or emit radiation energy in discrete packets proportional to the frequency of the radiation: energy **quanta**

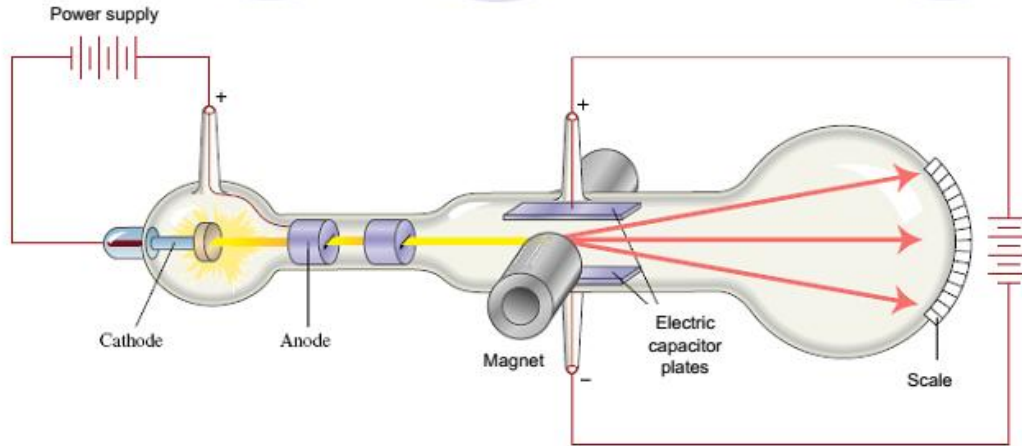


Albert Einstein

According to A. Einstein, Planck's quantization arises from the granular nature of light: light is made up of **photons**

1898年

Atoms are no longer indivisible !

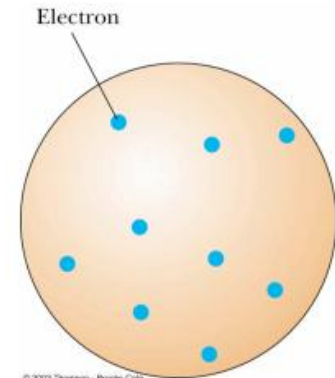


John Joseph Thomson

J.J. Thomson measures the charge/mass ratio of the charged particles

1898: J.J. Thomson concludes that what he calls 'corpuscles' (= electrons) are the constituents of atoms

'plum pudding' model



1895年

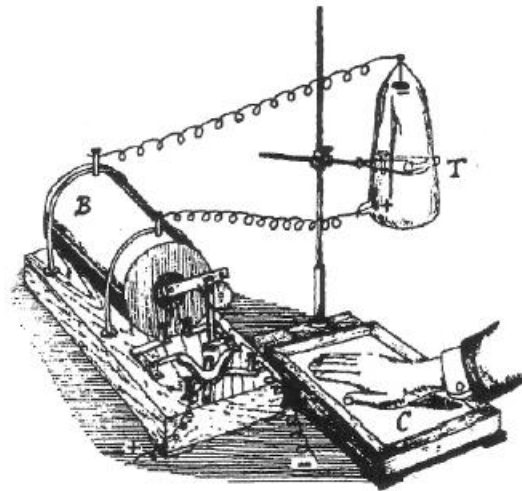
From cathode rays to X rays

1895 W. Röntgen

discovery of X rays



Wilhelm Röntgen



EJC2011, A. Lopez-Martens

W. Röntgen receives the 1st Nobel Prize in Physics in 1901



1896年

From X rays to Uranium rays

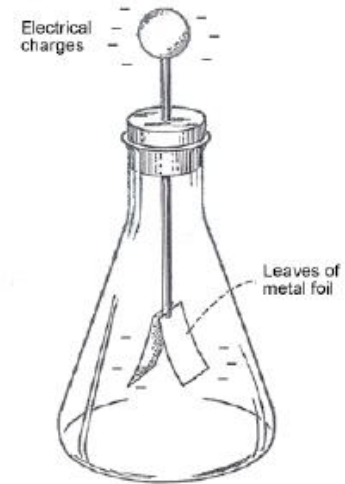
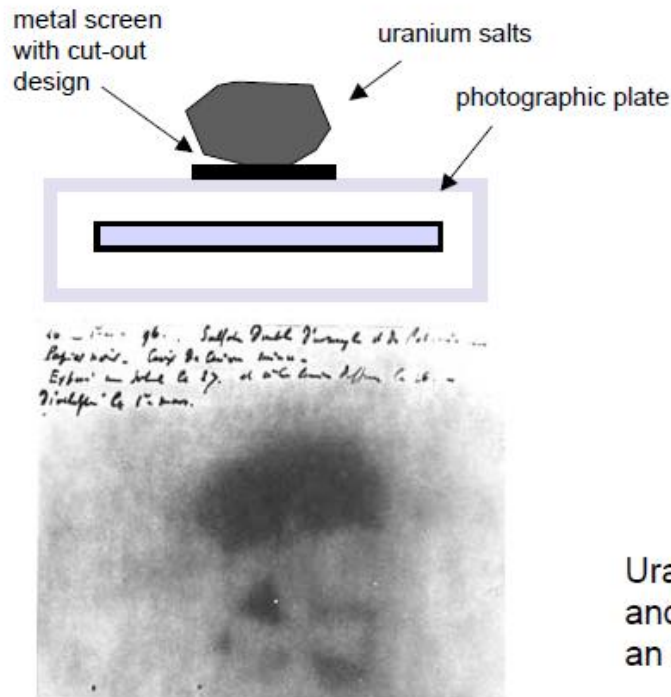
1896 H. Becquerel

discovery of a new kind of radiation emitted by Uranium



Henri Becquerel

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Uranium rays ionize the air and cause the discharge of an electroscope

Nobel Prize in Physics (1903)

From Uranium rays to radioactivity

1898年

1898 Marie & Pierre Curie

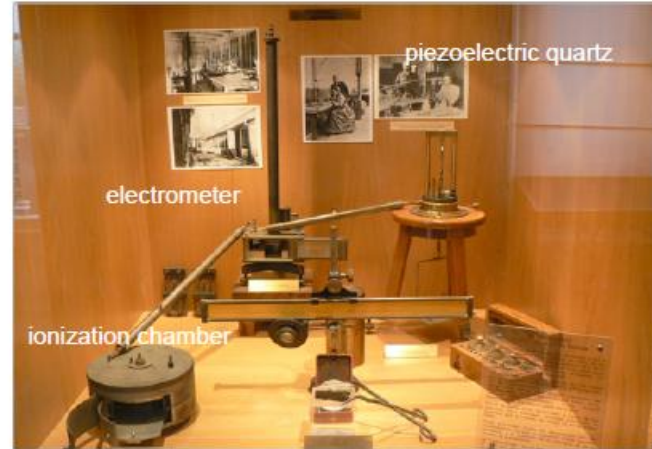
extraction of polonium and radium



Marie Curie



Pierre Curie



M. Curie calls the radiation:
'radioactivity'

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Laboratory at the Ecole de Physique et Chimie industrielle de Paris

因为天然放射性方面的贡献，居里夫妇跟贝克勒尔同一年分享一半 Nobel Prize in Physics (1903)

居里夫人，1911年，因发现元素钋和镭再次获得诺贝尔化学奖

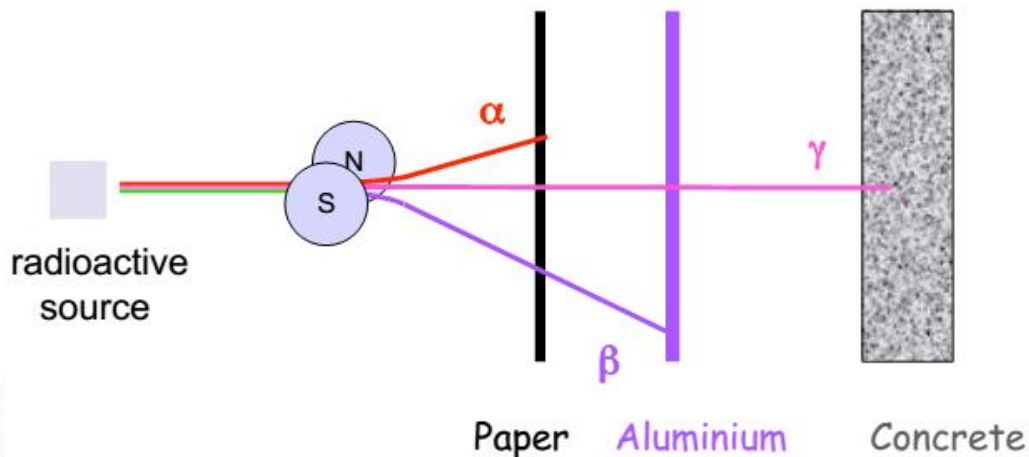
Radioactivity is manifold

1898 E. Rutherford

alpha, beta radiation

1900 P. Villard

gamma radiation



α = hélium ion He^{2+}

β = high energy e^-

γ = photons - like X rays

1908年Noble化学奖，这个工作是得化学Noble的主要基础，通过alpha decay后，元素蜕变别的元素.

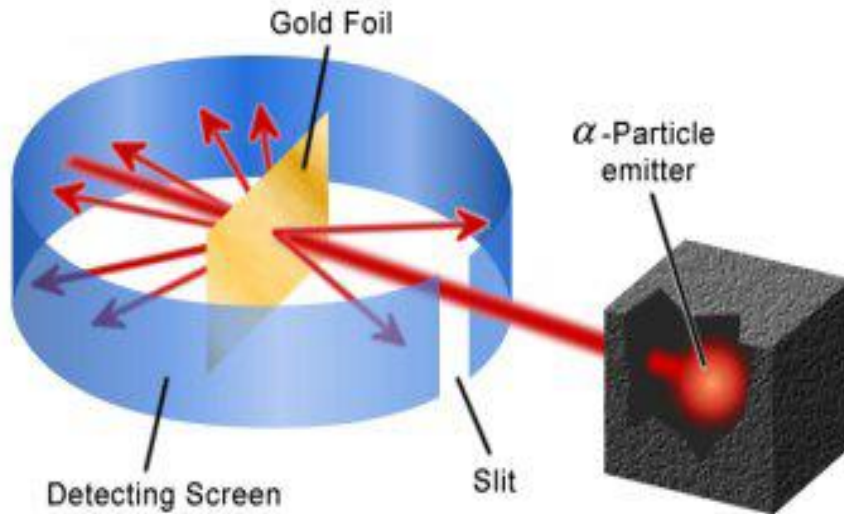


Ernest Rutherford



Paul Villard

- 1908: H. Geiger 和 E. Marsden 在研究 α 粒子散射实验中发现，约每 8000 个 α 粒子，就有一个被大角度散射。



E. Rutherford
(1871 - 1937)

H. Geiger, *Proceedings of the Royal Society of London A* 81, 174 (1908).

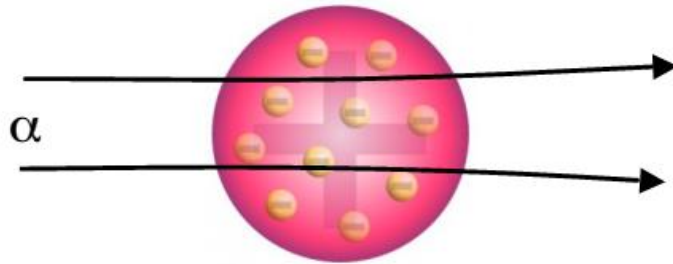
E. Rutherford, *Philosophical Magazine* 21, 669 (1911).

“It was almost as incredible as if you fired a 15-inch shell at a piece of tissue paper and it came back and hit you!”

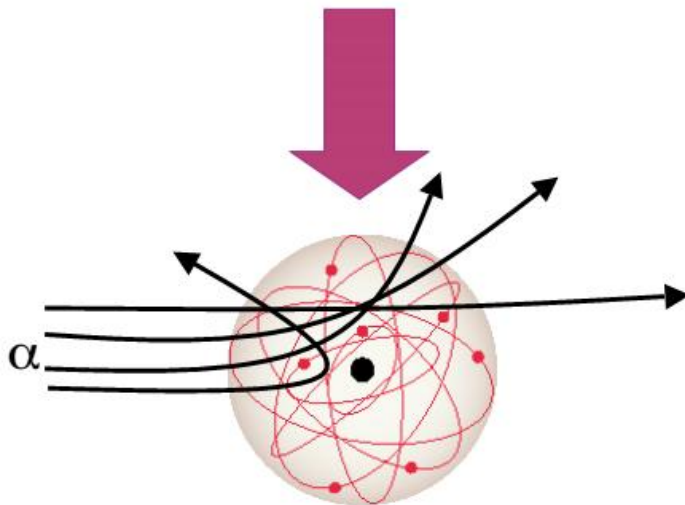
1911年，形成有核原子模型概念

The nucleus is born !

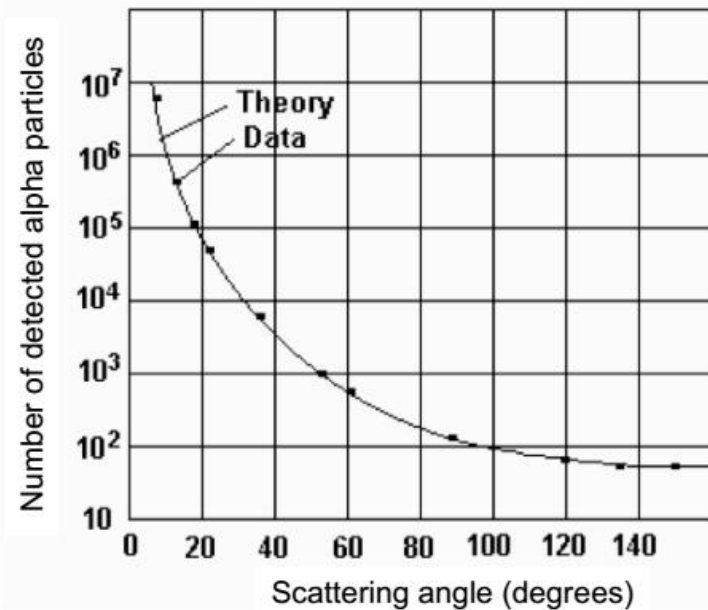
'plum pudding' model



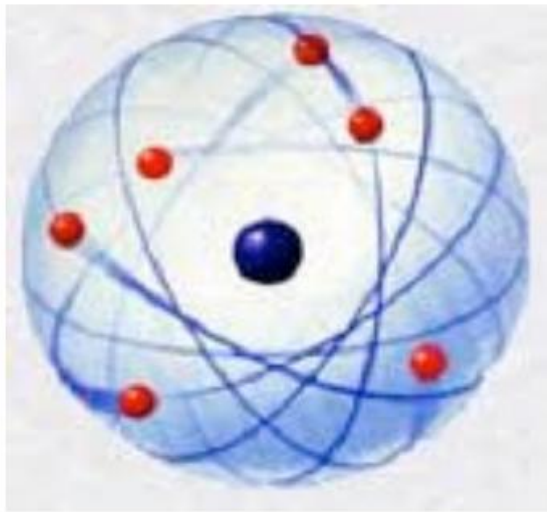
Philosophical Magazine Series 6,
vol. 21 May 1911, p. 669-688



Nuclear model



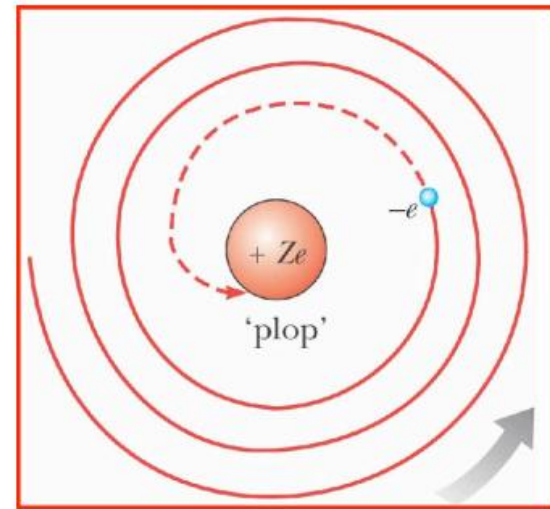
Consequences of Rutherford's model



10^{-14} m

10^{-10} m

Matter is empty !



The atom is unstable !

1913年波尔原子模型

Bohr solves the problem with Planck's quanta

- angular momentum is quantized \Leftrightarrow only certain orbits of energy E_n ($n=1,2,3,\dots$) are allowed
- n is the principal quantum number
- electronic jumps from one orbit to another are accompanied by the emission (or absorption) of a photon of given wavelength



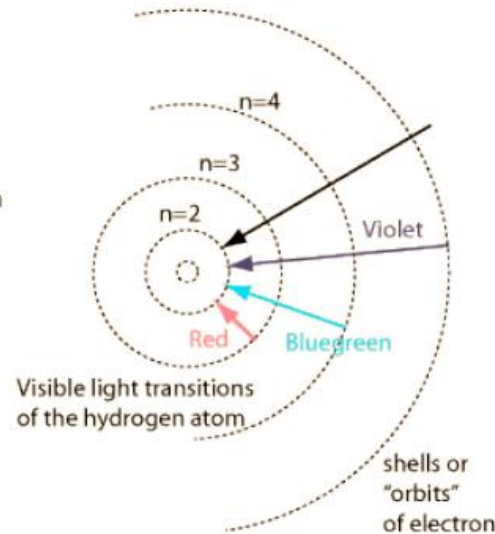
Niels Bohr

Philosophical Magazine 26, 1-25 (1913)

$$E_n \propto -Q^2/n^2$$

n=4
n=3
n=2
Energy Levels of Hydrogen

n=1

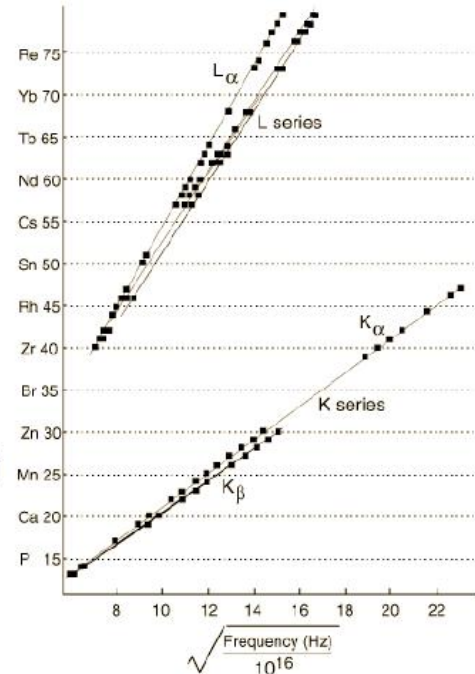
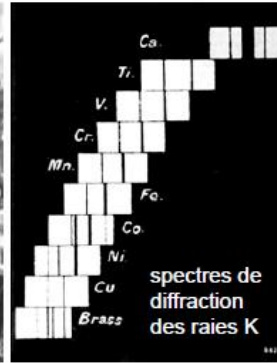


1922年 诺贝尔物理学奖

在玻尔原子模型基础上，莫塞莱提出了原子序数的概念，解释元素化学性质由核外的电子壳层排序确定的，推动了原子物理的发展。

一战爆发以后，莫塞莱辞去了他在牛津大学的工作，入伍成为一位工程兵军官。1915年，莫塞莱阵亡，年仅27岁。如果没阵亡，很可能诺贝尔奖。因为他的阵亡，英国政府制定了新的参战资格政策。

The periodic table finds its thread



1914 Henry Moseley

Measurement of the high frequency spectra of many elements

⇒ atomic number Z = charge of the nucleus

⇒ regularities of the periodic table explained in terms of filling of electronic shells

EJC2011, A. Lopez-Martens

Adapted from Moseley's original data (H. G. J. Moseley, Philos. Mag. (6) 27:703, 1914)

The alchemists of the 20th century

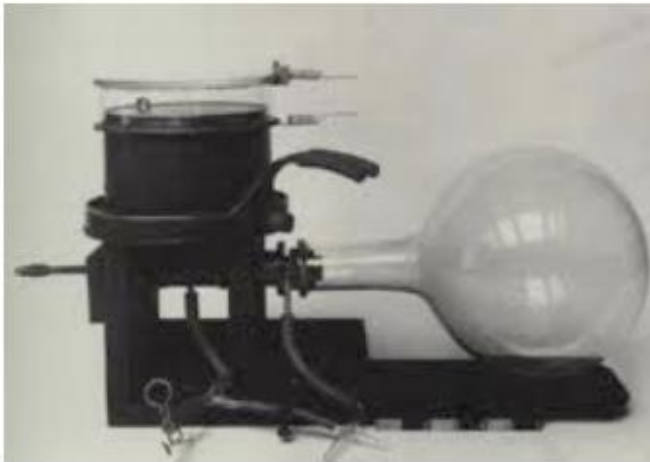
1919 E. Rutherford

1st artificial transmutation

$\alpha + \text{Nitrogen} \rightarrow \text{Hydrogen}$

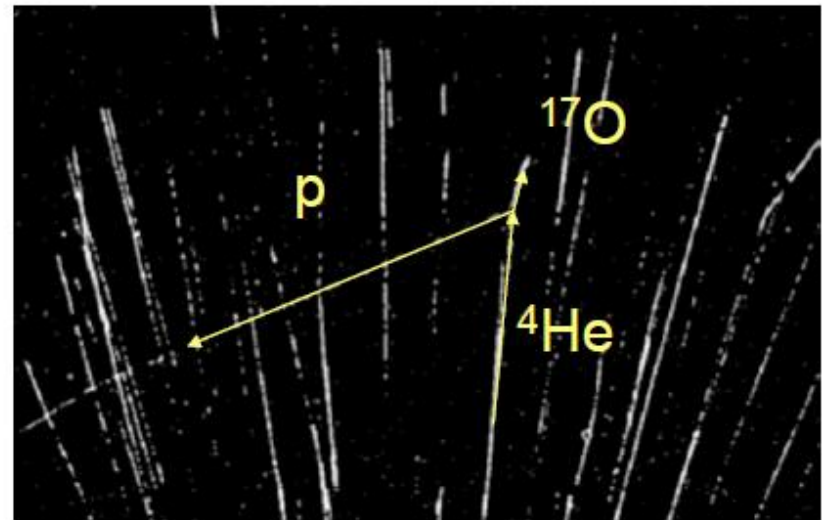
Rutherford calls H^+ proton 发现质子

1924 P. Blackett visualizes the transmutation $\alpha + {}^{14}\text{N} \rightarrow {}^{17}\text{F}^* \rightarrow {}^{17}\text{O} + \text{p}$



EJC2011, A. Lopez-Martens

Cloud chamber (C.T.R. Wilson, 1912)



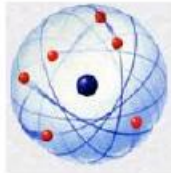
P. Blackett 1948 云雾室实验工作得 Nobel Prize

核物理早期二个代表性人物

100 years ago...

"The scattering of α and β particles by matter and the structure of the atom"

Philosophical Magazine Series 6, vol. 21 May 1911, p. 669-688



EJC2011, A. Lopez-Martens

Rutherford被誉为核物理之父

The Nobel Prize in Chemistry 1911 was awarded to Marie Curie *"in recognition of her services to the advancement of chemistry by the discovery of the elements radium and polonium, by the isolation of radium and the study of the nature and compounds of this remarkable element"*.



1898年：英国科学家 汤姆逊 发现电子

1911年：卢瑟福 α 背散射实验 原子模型

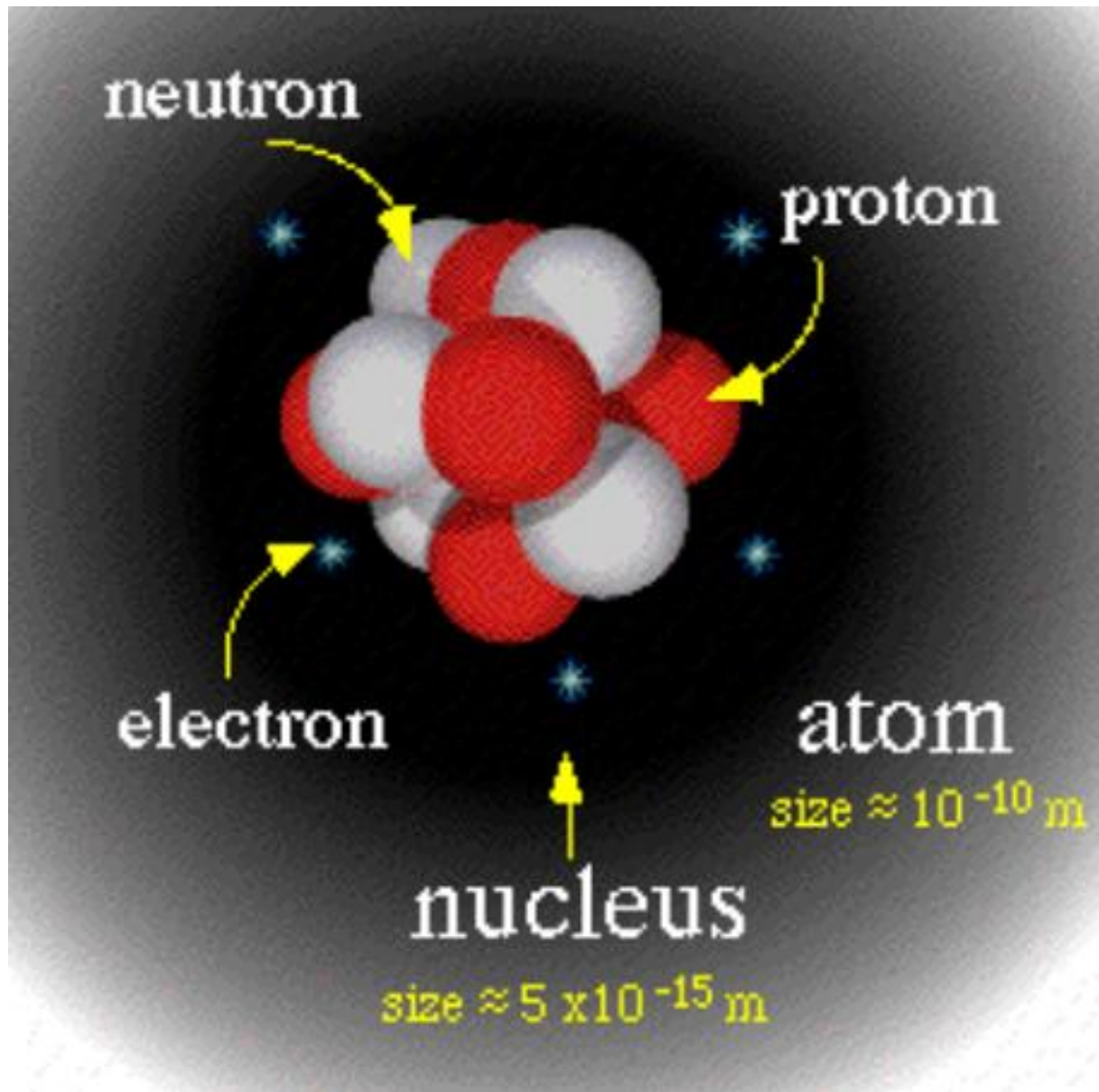
1913年：N.玻尔 量子论 建立玻尔原子模型

1919年 (实验1917, 发表1919)：卢瑟福 发现质子

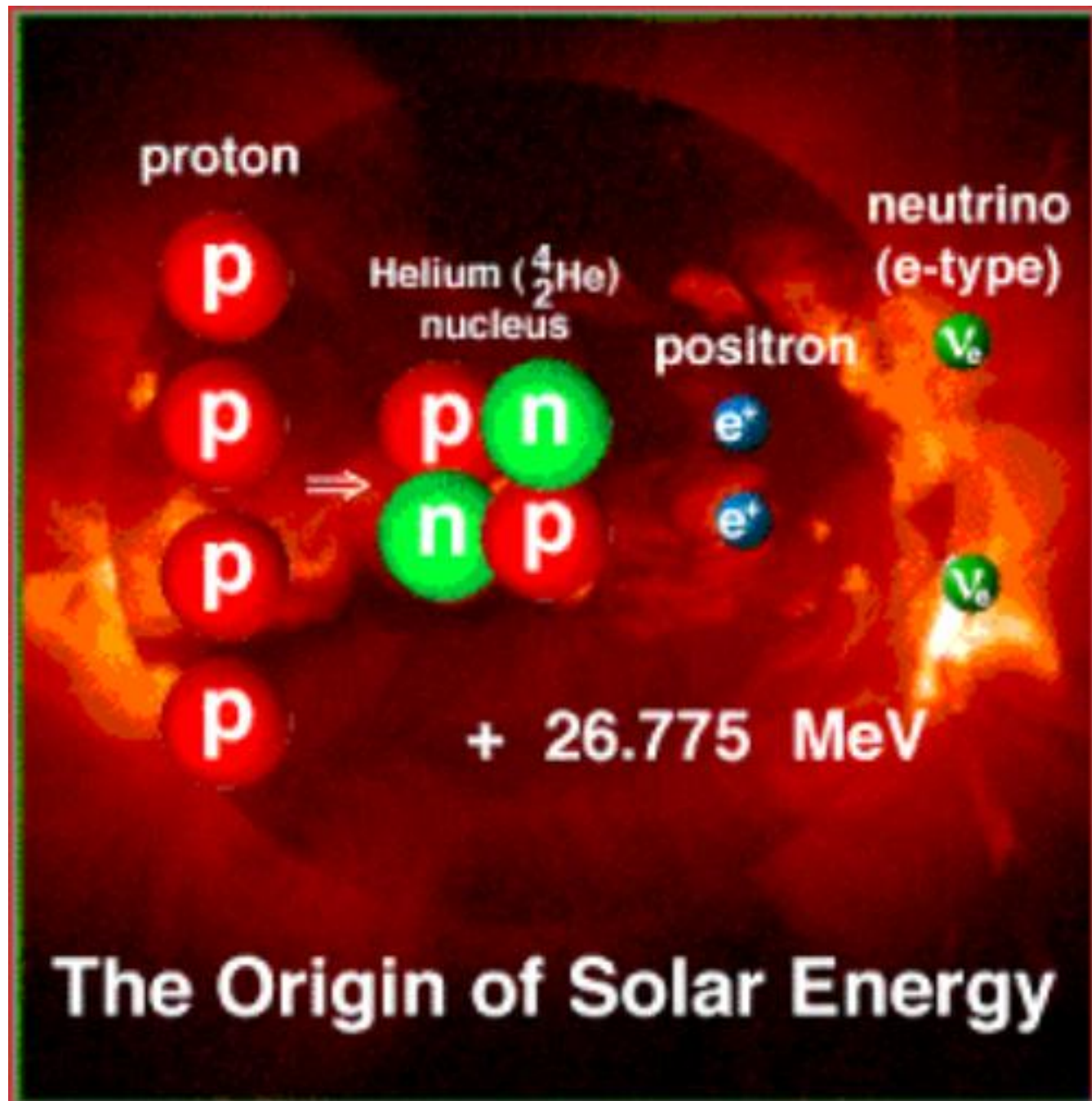
1932年：查德威克 发现中子

原子和原子核模型的建立

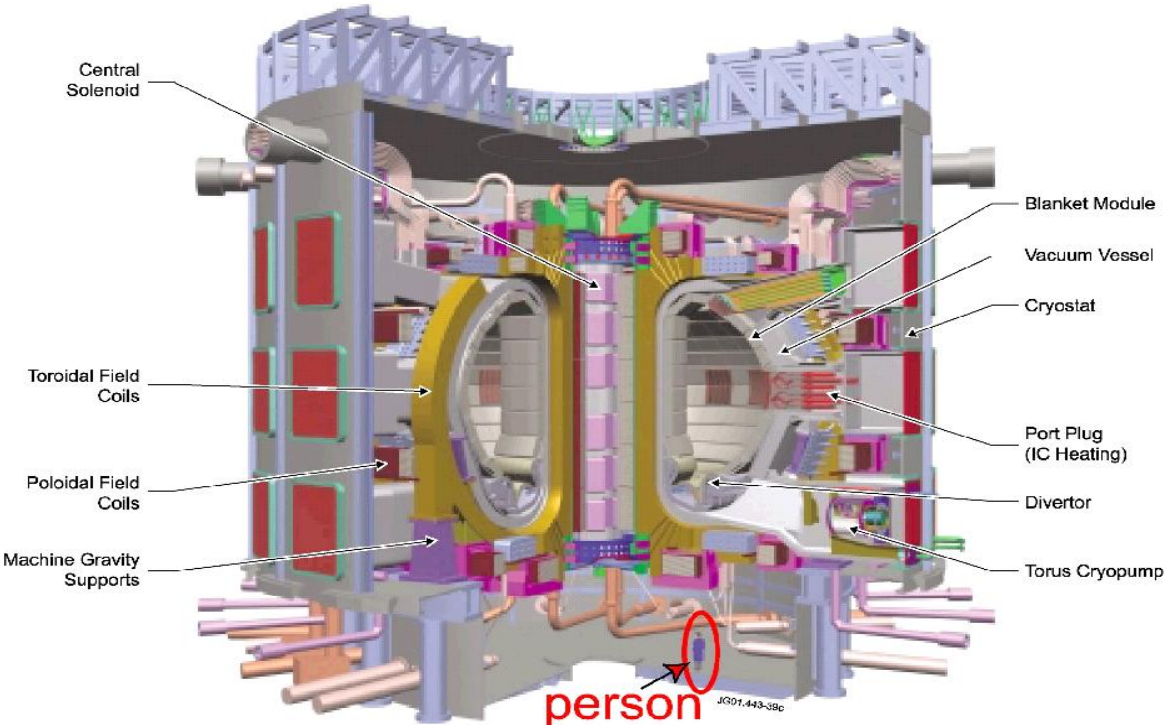
原子核结构



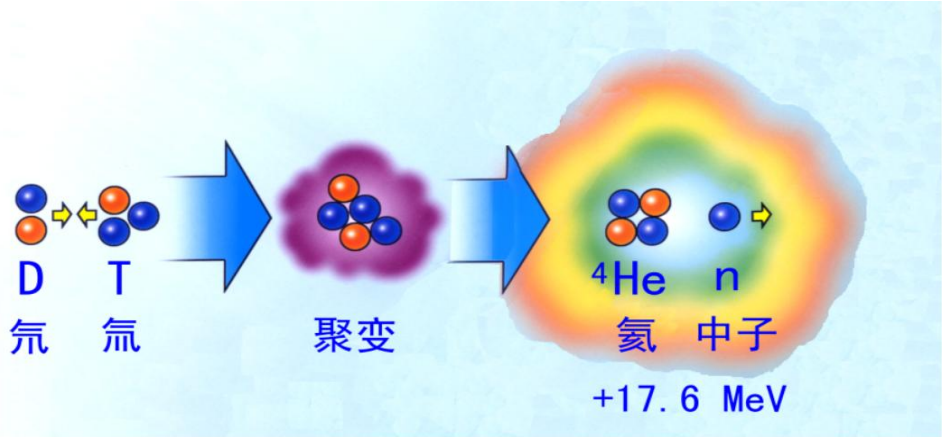
太阳能源



ITER装置



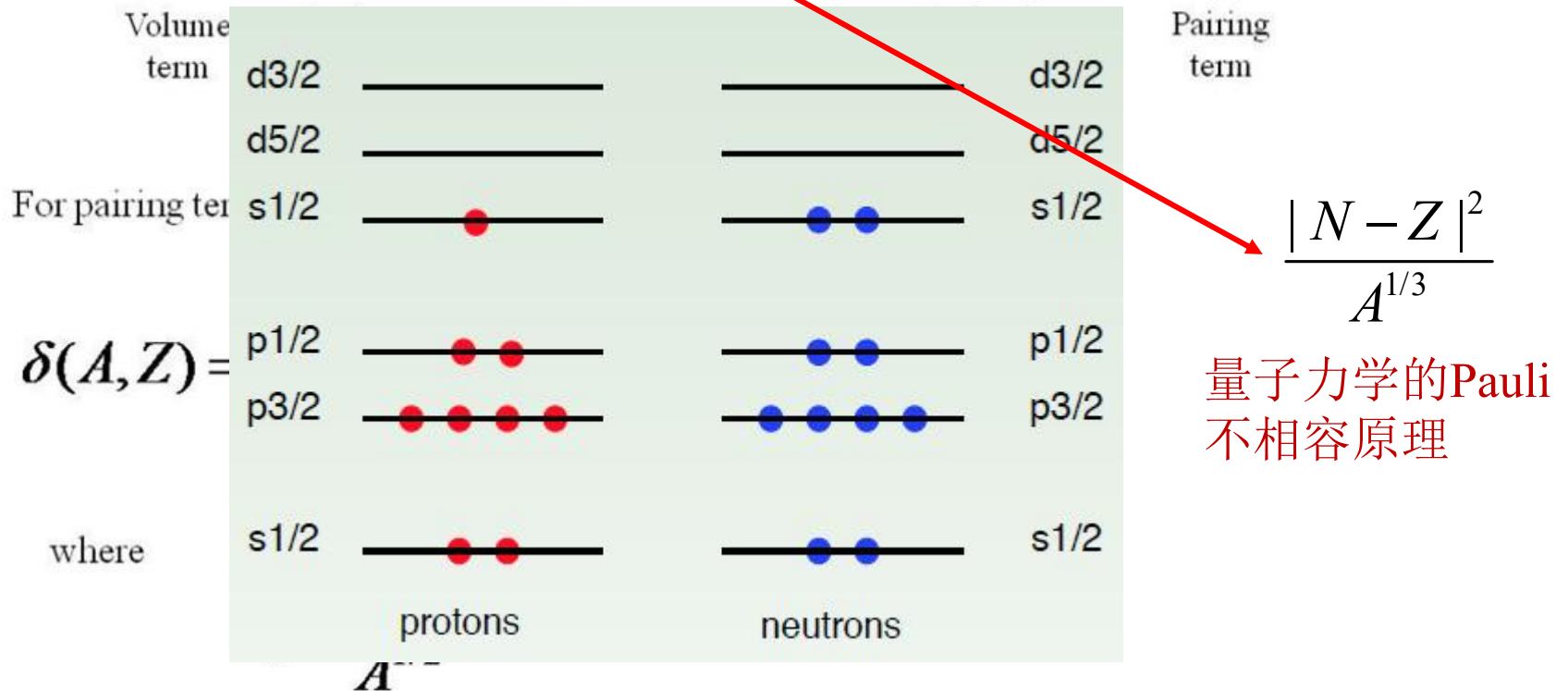
13.11.01



核物理发展的几个里程碑

液滴模型（Gamow提出概念，1935年由Weizsacker数学化，Bethe和N. Bohr贡献）

$$E_B = a_V A - a_S A^{2/3} - a_A \frac{(A-2Z)^2}{A^{1/3}} - a_C \frac{Z(Z-1)}{A^{1/3}} + \delta(A, Z)$$



裂变发现，N. Bohr用液滴模型，马上意识到裂变产生能量！

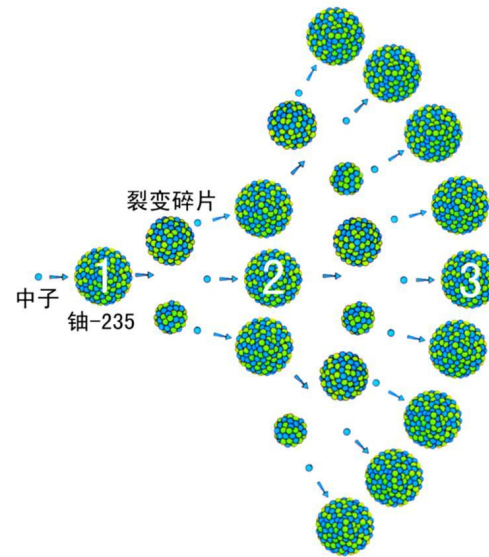
Q1: 对称能（Asymmetry or Wigner term)物理原因是什么？

核裂变的发现 (1938年: 哈恩, 迈特纳)



$$B(A, Z) = a_{vol} A - a_{surf} A^{2/3} - a_{comb} Z(Z - 1) A^{-1/3} - a_{sym} (A - 2Z)^2 A^{-1}$$

裂变: 核能



费米: 第一个反应堆

裂变链式反应

Nuclear Shell Model

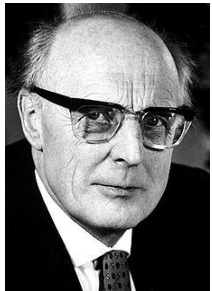
(概念1932年已经提出, 1949年发展, 1963 Nobel prize)



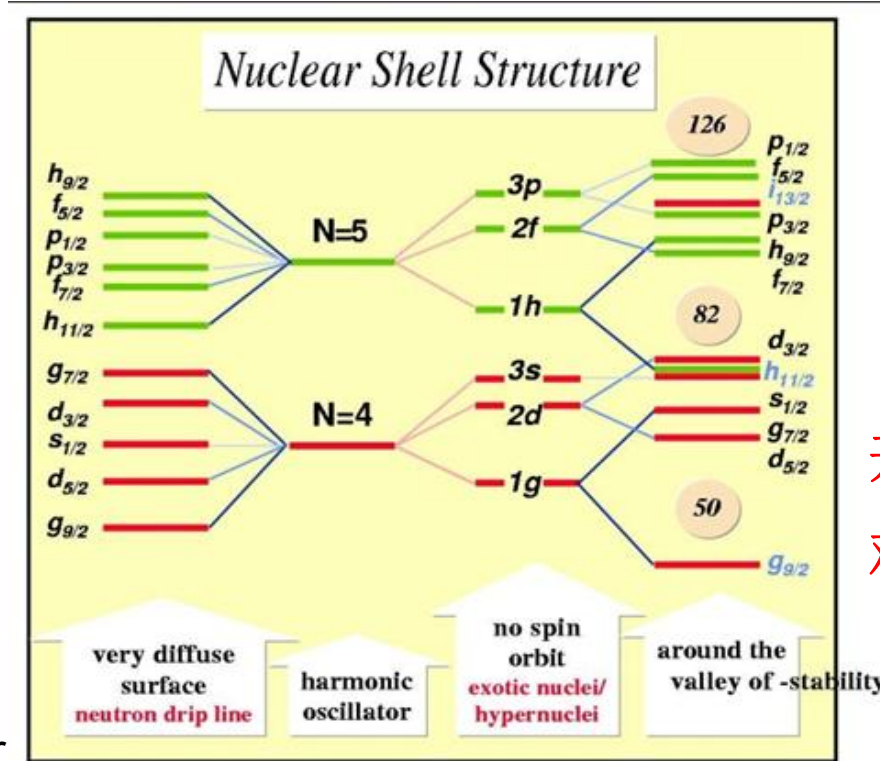
Eugene Paul Wigner



Maria Goeppert-Mayer



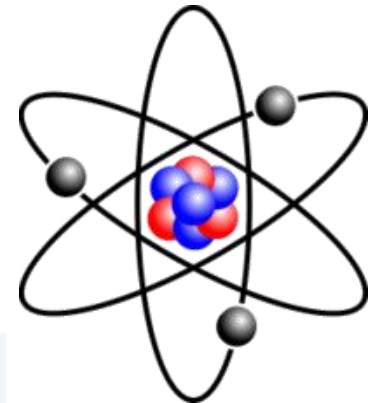
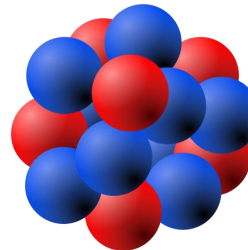
J. Hans D. Jensen



Spin-orbit coupling

幻数

开启了核结构的微观量子理论研究



从Liquid drop model: 原子核总是球形的!

高速转动原子核

The Nobel Prize in Physics 1975



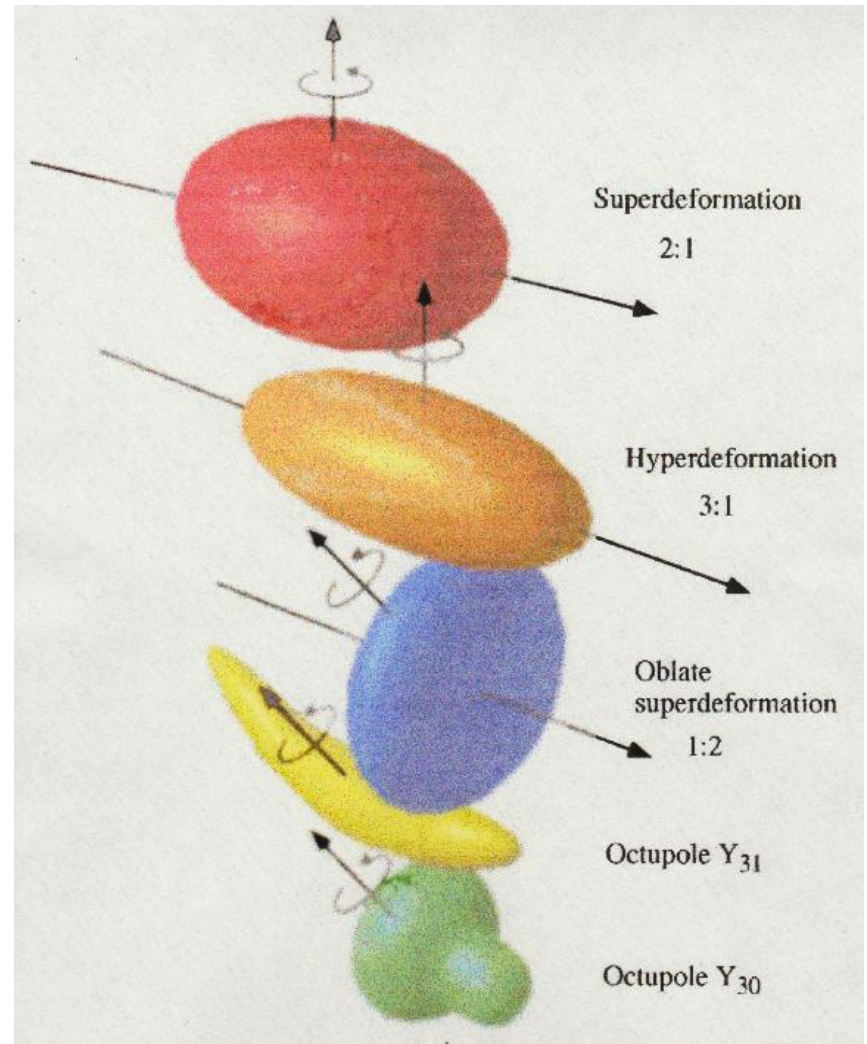
Aage Niels Bohr



Ben Roy Mottelson



Leo James Rainwater



The Nobel Prize in Physics 1975

The Royal Swedish Academy of Sciences has decided to award the Nobel Prize in Physics for 1975 in equal shares to Professor **Aage Bohr**, Denmark, to Professor **Ben Mottelson**, Denmark and to Professor **James Rainwater**, USA, for **the discovery of the connection between collective motion and particle motion in atomic nuclei and the development of the theory of the structure of the atomic nucleus based on this connection.**



Aage Niels Bohr



Ben Roy Mottelson



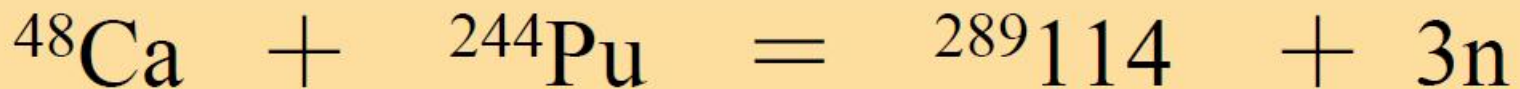
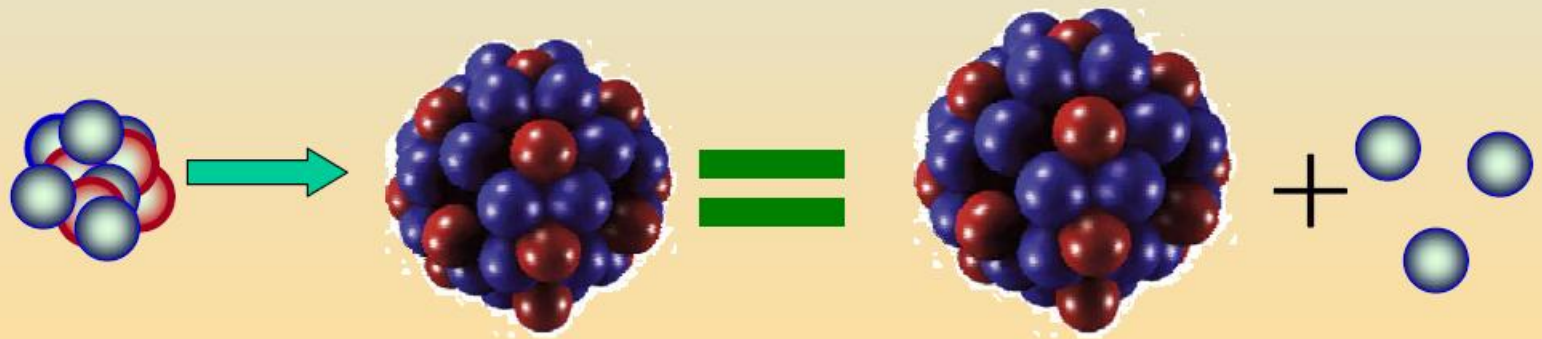
Leo James Rainwater

III. 目前的核物理前沿

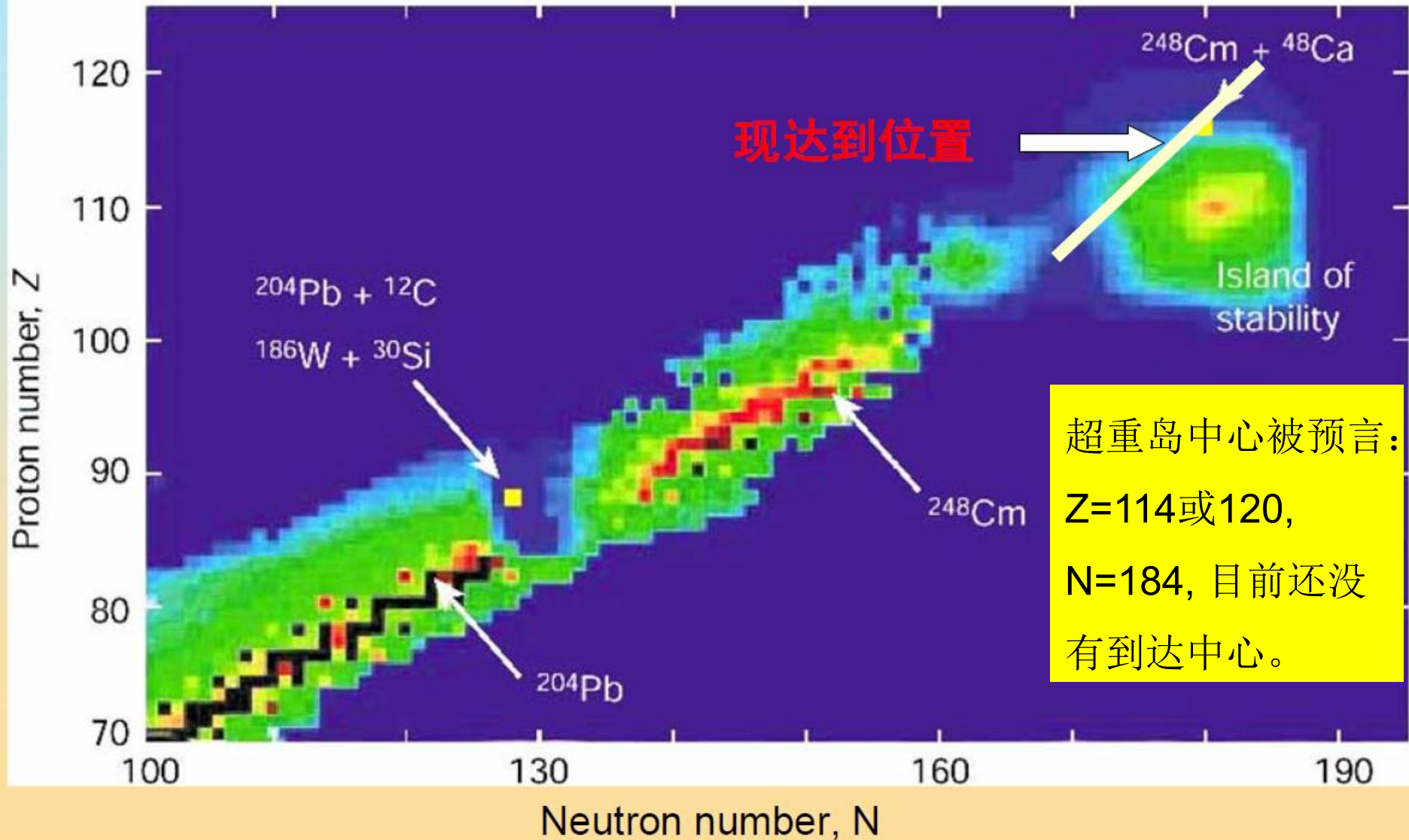
1. 超重元素
2. 滴线区原子核
3. 与基本对称性相关的精确测量
4. 新理论手段
5. 实验核天体物理

超重元素合成的途径

熔合反应——利用一定能量的炮弹原子核轰击靶原子核，两者熔合在一起，形成一个新的原子，它的原子序数为弹核与靶核之和。新原子核有一定的激发能，一般通过发射中子退激发。

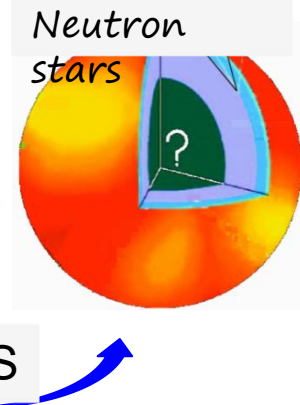
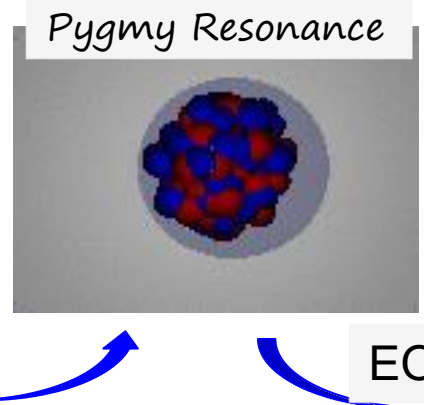
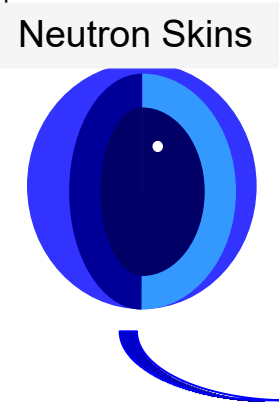
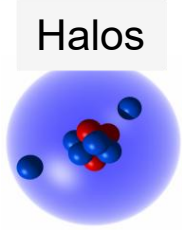
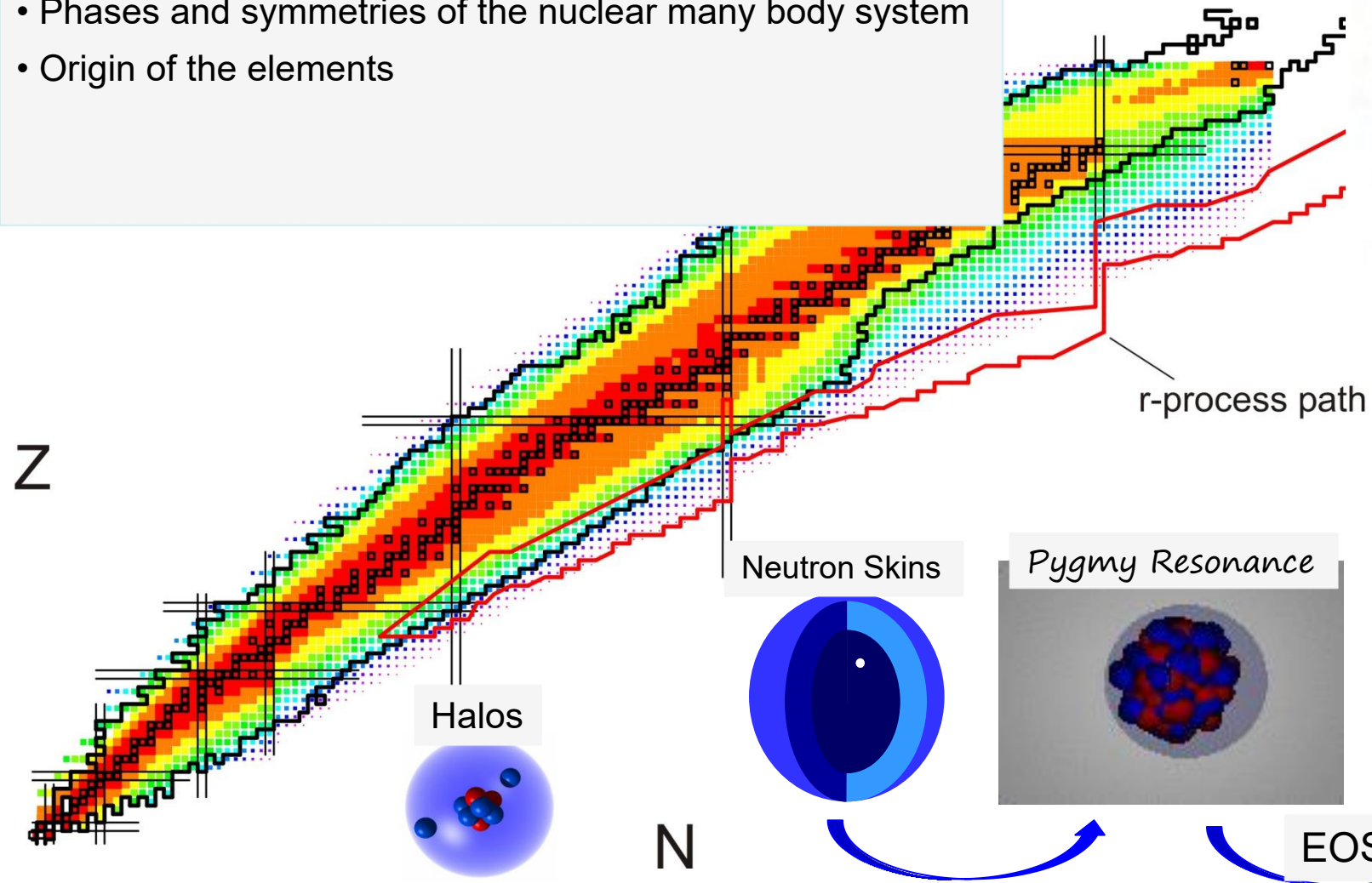


• 超重元素的合成



Challenging physics

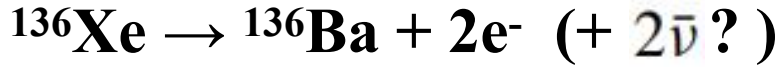
- Quest for the limits of existence
- Halos, Open Quantum Systems, Few Body Correlations
- Changing shell structure far away from stability
- Skins, new collective modes, nuclear matter, neutron stars
- Phases and symmetries of the nuclear many body system
- Origin of the elements



EOS

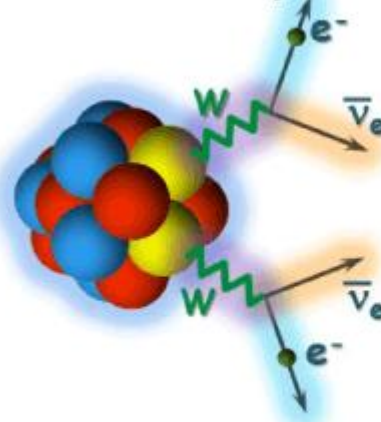
无中微子双 β 衰变 ($0\nu\beta\beta$)

Neutrino is its own antiparticle?
(Majorana Fermion)

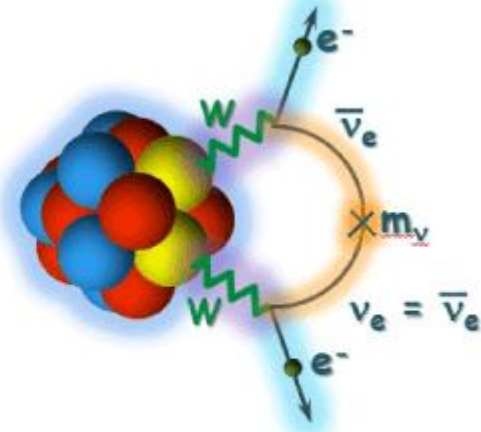


(利用精确的能量守恒关系)

[Double beta decay]



Double beta decay
which emits anti-neutrinos



Neutrinoless
double beta decay

预言衰变寿命: $[T_{1/2}^{0\nu}]^{-1} = G^{0\nu} |M^{0\nu}|^2 |f(m_i, U_{ei})|^2$

$G^{0\nu}$ phase-space factor (or kinematic factor); $f(m_i, U_{ei}) = g_A^2 \frac{\langle m_\nu \rangle}{m_e}$

$M^{0\nu}$ nuclear matrix element

(核物理第一性原理计算, 非常依赖于母核、子核波函数)

物理可观测量:

Free space \Rightarrow effective space, 利用场论重整化方法(Goldstone diagrams)

暗物质粒子 (WIMPs) 与原子核散射: 暗物质直接测量

Step 1: 场论扩展 $\mathcal{L}_{\text{eff}} = \mathcal{L}_{\pi\pi} + \mathcal{L}_{\pi N} + \mathcal{L}_{NN} + \dots$

WIMP质量: 1 — 100 GeV

PHYSICAL REVIEW LETTERS **128**, 072502 (2022)

Ab Initio Structure Factors for Spin-Dependent Dark Matter Direct Detection

B. S. Hu^{1,*}, J. Padua-Argüelles^{1,2,†}, S. Leutheusser^{1,3,‡}, T. Miyagi^{1,§}, S. R. Stroberg^{4,||,¶} and J. D. Holt^{1,5,**}

¹TRIUMF, 4004 Wesbrook Mall, Vancouver, British Columbia V6T 2A3, Canada

²Perimeter Institute, 31 Caroline Street North, Waterloo, Ontario N2L 2Y5, Canada

³Center for Theoretical Physics, Massachusetts Institute of Technology, Cambridge, Massachusetts 02139, USA

⁴Department of Physics, University of Washington, Seattle, Washington 98195, USA

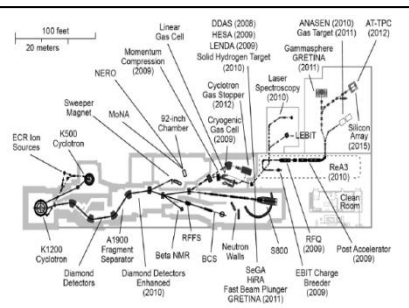
⁵Department of Physics, McGill University, Montréal, Quebec City H3A 2T8, Canada

目标 — 为暗物质头粒且按测量提供:

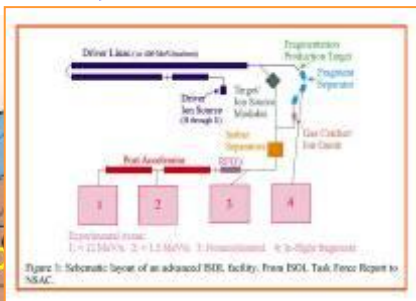
- 1) 暗物质粒子与原子核散射的最佳候选核
- 2) 散射截面
- 3) 最佳能区

世界上新建、在建、拟建放射性核大装置

MSU-NSCL



美国FRIB

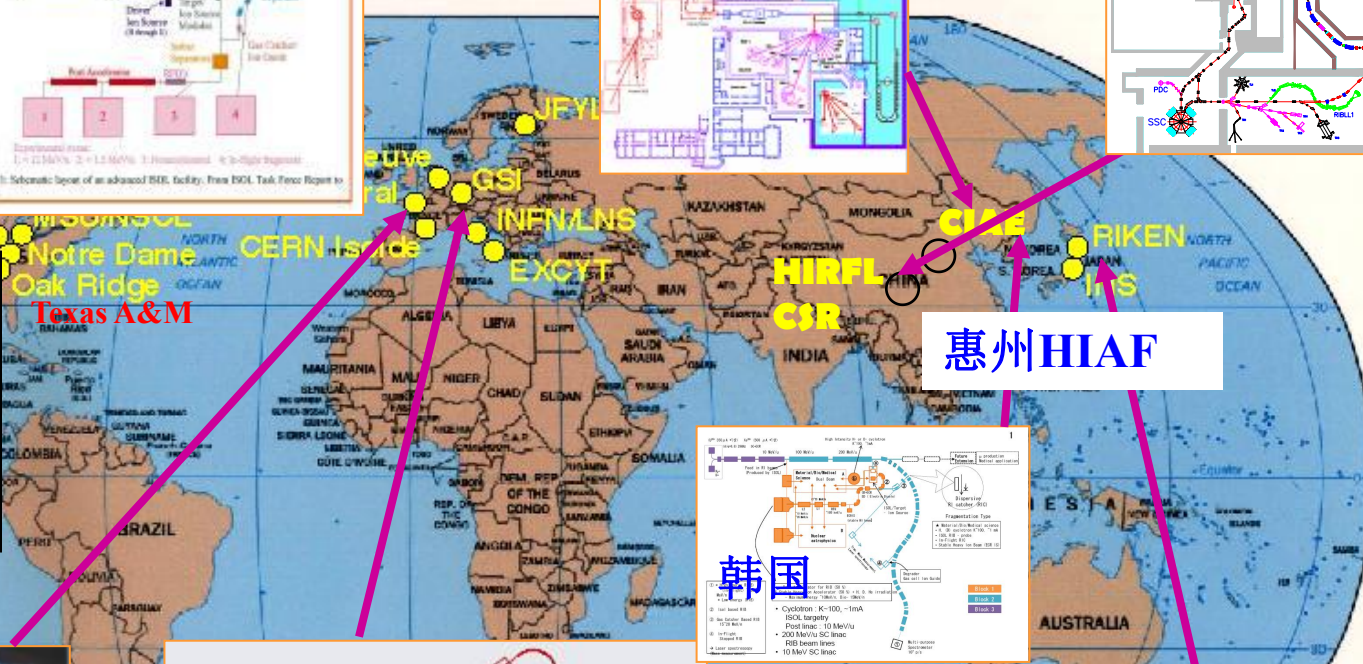
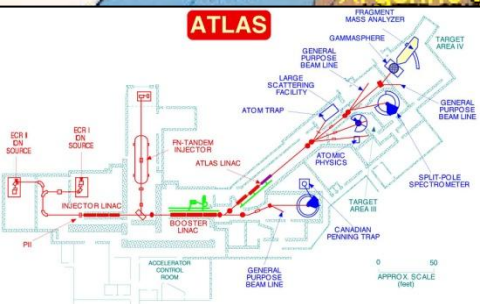
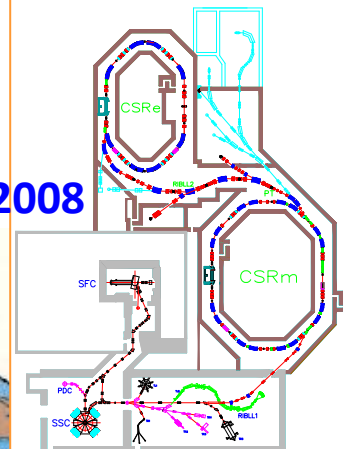


中国BRIF 2014

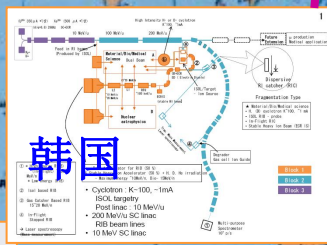


中国

CSR 2008

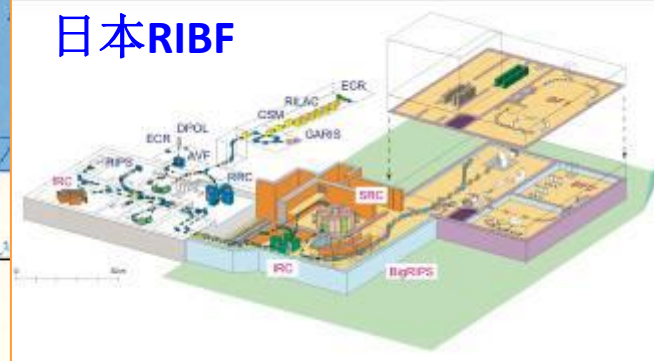


惠州HIAF



韩国

日本RIBF

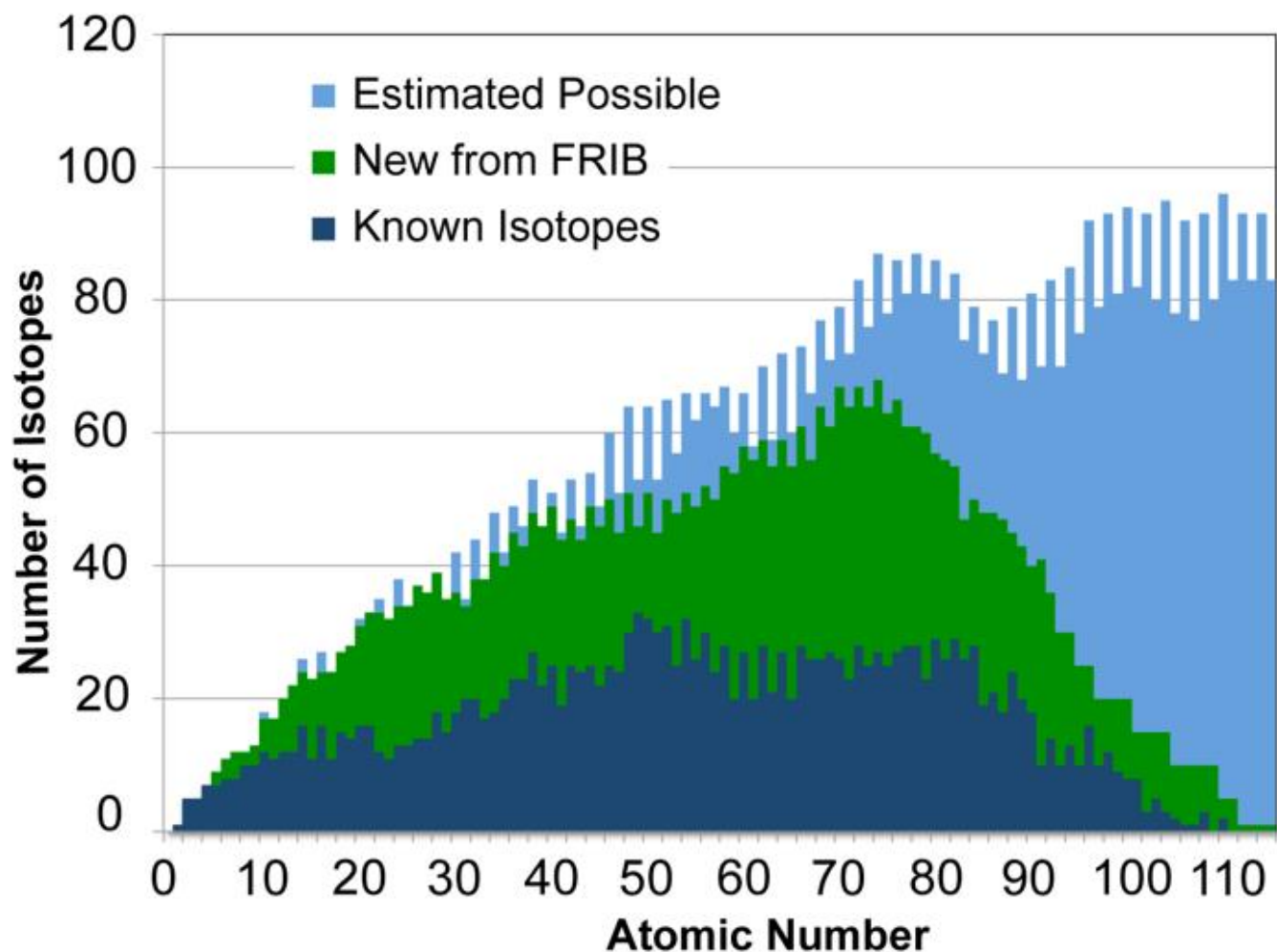


法国SPIRAL-II 2



德国FAIR

The Number of Isotopes Available for Study



- Estimated Possible: Erler, Birge, Kortelainen, Nazarewicz, Olsen, Stoitsov, to be published, based on a study of EDF models
- “Known” defined as isotopes with at least one excited state known
- Below $Z=90$ FRIB will be able to make >80% of all possible isotopes

核物理的新理论方法

核理论的二个最基本问题

1. 核力 (Nuclear Force, 强相互作用)
2. 强关联量子多体体系的求解 (many-body problems)

$$\hat{H}_{int} = \sum_{i < j}^A \frac{(\vec{p}_i - \vec{p}_j)^2}{2mA} + \sum_{i < j}^A V_{NN,ij} + \sum_{i < j < k}^A V_{NNN,ijk}$$

写出来难，处理起来难

最好从QCD出发，但目前还不是很成功...

Physics World公布2022年十大年度突破性成果



01 迎来超冷化学的新时代

02 观测到理论预言的四中子状态

两院院士评选“2022年中国/世界十大科技进展新闻”揭晓

中国科学报 2023-01-12 11:01 发表于北京

2022年世界十大科技进展新闻

01 首个完整人类基因组序列公布



10 科学家发现“四中子态”存在最明确证据

Physicists Observe Elusive Four-Neutron System: Tetraneutron

Jun 23, 2022 by Enrico de Lazaro

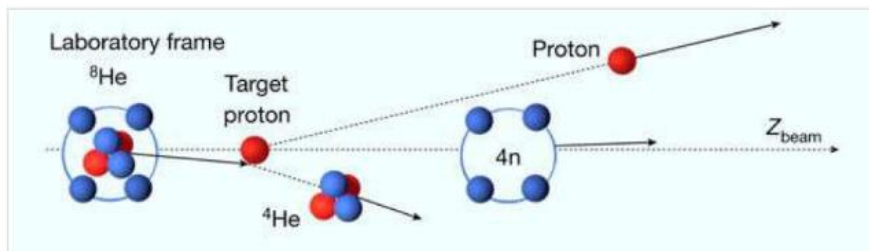
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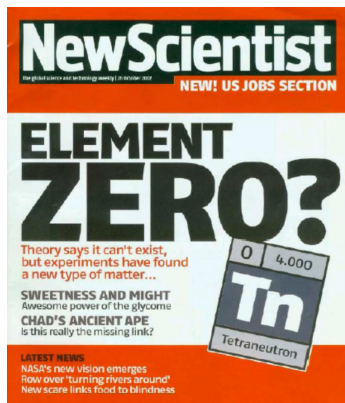
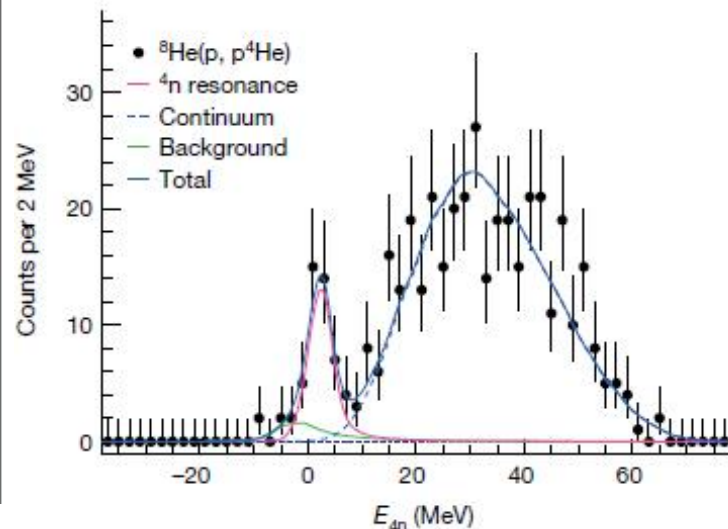
Physicists using the Superconducting Analyzer for Multi-particles from Radio Isotope Beams (SAMURAI) in Japan have experimentally observed a resonance-like structure consistent with a **tetraneutron state** after 60 years of experimental attempts to clarify its existence.



四中子共振态 (Tetraneutron)

(大概60年前, 猜想)

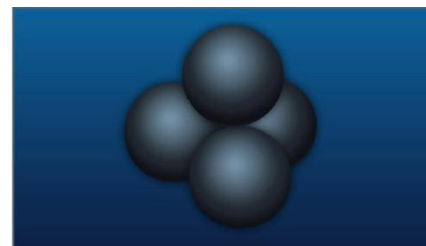
M. Duer *et al.*, Nature 606, 678 (2022)



ScienceNews
INDEPENDENT JOURNALISM SINCE 1921

NEWS PHYSICS

Physicists may have finally spotted elusive clusters of four neutrons
If confirmed, 'tetraneutrons' could help scientists better understand nuclear forces



Four neutrons may form a short-lived agglomeration called a tetraneutron (illustrated).
SONJA BATTENBERG/TUM

By Emily Conover
JUNE 22, 2022 AT 11:00 AM

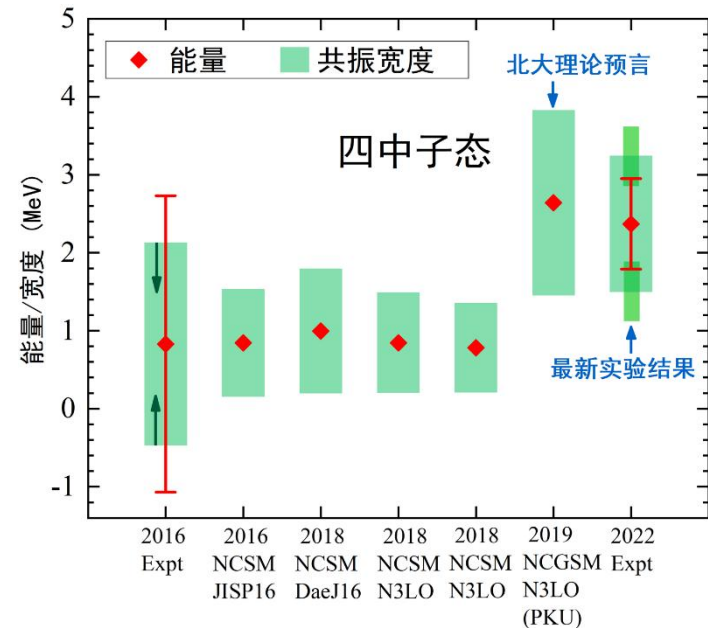
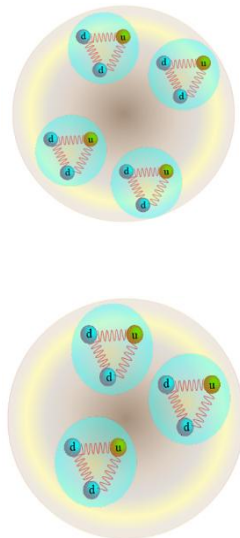
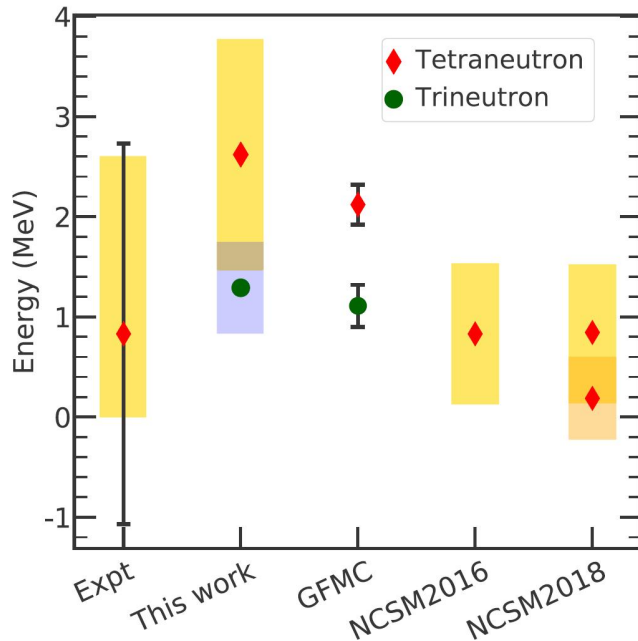
Ab initio no-core Gamow shell-model calculations of multineutron systems

J. G. Li,¹ N. Michel ,^{2,3} B. S. Hu ,¹ W. Zuo,^{2,3} and F. R. Xu ^{1,*}

¹*School of Physics, and State Key Laboratory of Nuclear Physics and Technology, Peking University, Beijing 100871, China*

²*Institute of Modern Physics, Chinese Academy of Sciences, Lanzhou 730000, China*

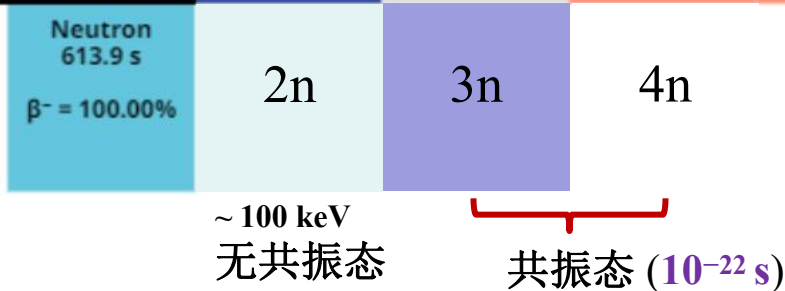
³*School of Nuclear Science and Technology, University of Chinese Academy of Sciences, Beijing 100049, China*



Z=1

1H STABLE 99.9885%	2H STABLE 0.0115%	3H 12.32 y $\beta^- = 100.00\%$	4H N = 100.00%	5H 5.7 mev 2N = 100.00%	6H 1.6 mev N = 100.00%	7H 500 ys 2N? 10^{-24} s
--------------------------	-------------------------	---------------------------------------	-------------------	-------------------------------	------------------------------	-------------------------------------

Z=0



发现4n体系的重要性:

- 1) 核力 (nn)
- 2) 纯中子物质

...

$0p_{1/2}$



$0p_{3/2}$



$0s_{1/2}$

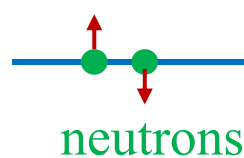


protons

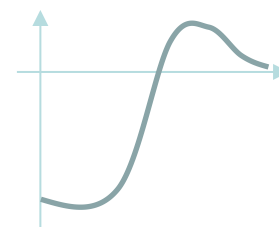
$0p_{1/2}$



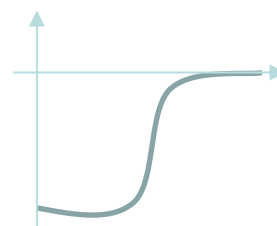
$0s_{1/2}$



neutrons



l=1离心位垒



l=0 无离心位垒

小结

历史：在量子力学建立初期，核物理起非常重要的作用

- 核物理还有许多事情要做！
- 新一代大型先进核设施建设将给我们带来空前的机遇，同时对理论是挑战！
- 还会有新的激动人心的新发现！

Brief History of Nuclear Physics

- 1896: discovery of radioactivity by Becquerel
- 1898: separation of Radium by Maria and Pierre Curie; discovery of α , β , γ rays
- 1911: nucleus as a central part of an atom – Rutherford
- 1913: Soddy and Richards elucidate the concept of nuclear mass: isotopes are born
- 1919: Rutherford carries out first transmutation ($\text{He} + \text{N} \rightarrow \text{p} + \text{O}$)
- 1923: Georg von Hevesy uses radioactive tracers in biology
- 1928: theory of alpha decay by Gamow
- 1929: cyclotron (Ernest Lawrence); Rasetti discovers spin $J=1$ for ^{14}N
- 1930: Pauli predicts neutrino; Dirac predicts antimatter
- 1932: discovery of the neutron by Chadwick; discovery of positrons by Anderson
- 1934: Fermi theory of beta decay; Baade and Zwicky predict neutron stars
- 1935: nuclear (strong) force through meson exchange – Yukawa
- 1936: John Lawrence treats leukemia with ^{32}P
- 1938: stars are powered by nuclear fusion (Gamow, von Weizsäcker, Bethe): pp, CNO
- 1939: nuclear fission (Hahn, Strassman, Meitner, Frisch); Bohr, Wheeler explain fission
- 1940: McMillan and Abelson produce a new element ($n + ^{238}\text{U} \rightarrow ^{239}\text{U} \rightarrow ^{239}\text{Np} \rightarrow ^{239}\text{Pu}$)
- 1942: first self-sustaining fission reaction (Fermi); Manhattan project (Oppenheimer)
- 1945: atomic bomb
- 1947: pi meson discovered in Bristol (by studying cosmic ray tracks)
- 1948: Big Bang nucleosynthesis (Alpher, Bethe, Gamow)
Electricity generated at the X-10 Graphite Reactor in Oak Ridge

1949: nuclear shell model (Mayer, Jensen)
1951: nuclear collective model (Bohr, Mottelson, Rainwater)
1952: hydrogen bomb (Teller, Ulam); Hoyle resonance predicted
1954: proton therapy at Berkeley
1956: experimental evidence for antineutrino (Reines, Cowan)
prediction and discovery of parity violation (Lee, Yang, Wu)
1957: stellar nucleosynthesis (Burbidge, Burbidge, Fowler, Hoyle)
1958: nuclear superconductivity (Bohr, Mottelson, Pines)
1961: first PET scan at Brookhaven
1964: quarks proposed (Gell-Mann, Zweig)
1967: discovery of neutron stars (Hewish, Shklovsky, Bell)
1969: intrinsic structure of the proton (SLAC)
1972: color charge and quantum chromodynamics (Fritsch, Gell-Mann)
1978: discovery of the gluon (DESY)
1982: chiral symmetry on the lattice (Ginsparg, Wilson)
1983: discovery of W and Z intermediate vector bosons (CERN)
1995: top quark discovered (Fermilab)
1999: discovery of particle stability of ^{31}F (RIKEN)
2001: neutrino oscillations (Super-Kamiokande, SNO)
2002: element $Z=118$ produced in Dubna
2005: quark-gluon liquid of very low viscosity discovered at RHIC
2008: discovery of ^{40}Mg at MSU <https://www.youtube.com/watch?v=tJsam4z715c>



CUSTIPEN

中美奇特核物理理论研究所

China-U.S. Theory Institute for Physics with Exotic Nuclei

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About CUSTIPEN

Purpose:

Deliver an international venue for research on the physics of nuclei during an era of experimental investigations on rare isotopes.

Location:

Peking University, Beijing, China

US Participation:


Provide travel and local support for visits of U.S.-based physicists to CUSTIPEN

China Participation:

Provide local support for China-based physicists at CUSTIPEN

Synopsis:

The China-U.S. Theory Institute for Physics with Exotic Nuclei (CUSTIPEN) has been established in order to facilitate collaborations between U.S. and Chinese scientists whose main research thrust is in the area of the physics of nuclei. U.S. participation in CUSTIPEN is in the form of travel grants and subsistence grants to those individuals who are interested in collaborating with Chinese scientists.



谢谢

2023.04.21 科大